



Clean Fuel Standard Discussion Paper

Ammonia (NH₃) as a Carbon-Free Fuel

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Ammonia as a Carbon-Free Fuel for Use in Transportation and Power Generation

The action plan lays out the specific commitments Canada is making from Ontario's point of view to meet its 15% overall greenhouse gas emissions reduction target by 2020. Emissions from total passenger transportation (cars, trucks, bus, rail and domestic aviation) have grown almost 15% since 1990, to 36 million tonnes of CO₂e, approximately 66% of Ontario's 2014 transportation emissions. This growth was driven by an increase in vehicle-kilometres travelled as well as a shift in the composition of the fleet from cars to sport-utility vehicles, pickups and minivans where the specific contributions of the vehicle types are shown in Fig 1.

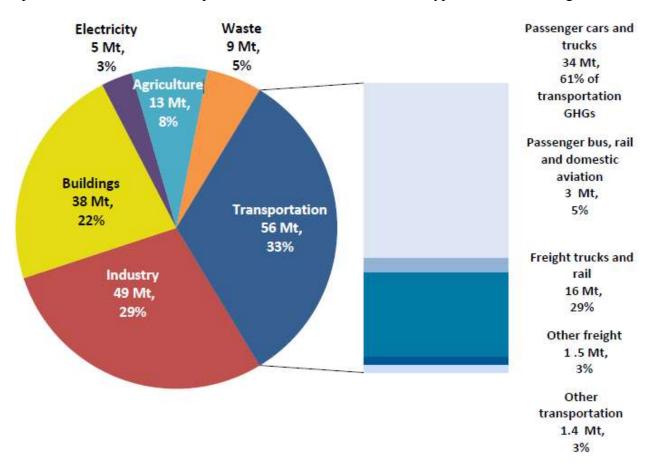


Fig. 1. Ontario's greenhouse gas emissions in 2014 (data from Ref. 1).

Here, it is important to note that under Environment and Climate Change Canada's economic sector categorization, most off-road transportation emissions are allocated to their host economic sectors and consequently are not included under transportation in Fig. 1. For example, emissions from diesel combustion in farm equipment are categorized under Agriculture. Therefore, in reality GHG emissions caused by the transportation vehicles are higher than 33% in Ontario.

Additionally, emissions from total freight transportation (trucks, rail and other) have increased more drastically over the period, rising 85% since 1990, to almost 18 million tonnes of CO₂e (approximately one third of Ontario's current transportation emissions). This was driven by a significant increase in the use of diesel-fuelled heavy-duty trucks, with additional kilometres travelled offsetting improvements in efficiency [1].

This brief report tries to address the following points:

1. Targets and blending requirements:

- a. Ontario's has existing content requirements for ethanol in gasoline. What minimum level of ethanol blending and GHG performance would help support the objectives of the RFS?
- b. Given Ontario's GHG reduction targets for 2030 and 2050, what factors should be considered in setting RFS targets post-2020?

2. Flexibility mechanisms:

- a. Should activities to lower the carbon intensity of other conventional transportation fuels be eligible for compliance purposes?
- b. Should investments in low-carbon transportation projects also be eligible for compliance purposes? If yes, what types of projects?

3. Assessing lifecycle emissions

a. Should an RFS consider impacts from indirect land-use changes (ILUC),7 even though science in this area continues to evolve? If so, how?

4. Transparency:

a. What measures can be taken to increase transparency and support business decision making under an RFS (e.g. an information registry, bulletins, guidance material)?

5. Others:

a. What other considerations should be included in the discussion?

AMMONIA FACTS

Ammonia is one of the largest synthesized industrial chemical in the world having over 200 million tonne per production per year.

Ammonia (NH₃):

- consists of one nitrogen atom from air separation and three hydrogen atoms from any conventional or renewable resources.
- is the second largest synthesized industrial chemical in the world.
- is a significant hydrogen carrier and transportation fuel that does not contain any carbon atoms and has a high hydrogen ratio.
- contains about 48% more hydrogen by volume than liquefied hydrogen.
- does not emit direct greenhouse gas emission during utilization
- can be used as solid and/or liquid for many purposes.
- can be stored and transported under relatively lower pressures.
- can be produced from various type of resources ranging from oil sands to renewables.
- is a suitable fuel to be transferred using steel pipelines with minor modifications which are currently used for natural gas and oil.
- can be used in all types of combustion engines, gas turbines, burners as a sustainable fuel with only small modifications and directly in fuel cells which is a very important advantage compared to other type of fuels.
- brings a non-centralized power generation via fuel cells, stationary generators, furnaces/boilers and enables smart grid applications.
- can be used as a refrigerant for cooling in the vehicle.

WHAT ARE THE USES OF AMMONIA?

Ammonia is considered a possible working fluid for thermodynamic cycles, working for refrigeration, heating, power or any mixture of those can be coupled with internal combustion engines, and using exhaust gasses to drive automotive absorption refrigeration system.

Ammonia has been recognized and employed as a leading refrigerant in the industrial sector due to its outstanding thermal properties, zero ozone depletion and global warming potential (GWP). Ammonia has the highest refrigerating effect per unit mass compared to all the refrigerants being used including the halocarbons. The remarkable advantages of ammonia over R-134a could be lower overall operating costs of ammonia systems, the flexibility in meeting complex and several refrigeration needs, and lower initial costs for numerous applications. Ammonia has better heat transfer properties than most of chemical refrigerants and consequently allows for the use of equipment with a smaller heat transfer area. Thereby plant construction cost will be lower. But as these properties also benefit the thermodynamic efficiency in the system, it also reduces the operating costs of the system. In many countries the cost of ammonia per mass is considerably lower than the cost of HFCs. This kind of advantage is even multiplied by the fact that ammonia has a lower density in liquid phase. Modern ammonia systems are fully contained closed-loop systems with fully integrated controls, which regulate pressures throughout the system. Ammonia is used as refrigerant highly in the refrigeration structures of the food industry like dairies, ice creams plants, frozen food production plants, cold storage warehouses, processors of fish, poultry and meat and a number of other uses.

It is also stimulating to note that NH₃ is a reduction agent for the NOx typically current in combustion releases. The reaction of NOx with ammonia over catalysts produces only steam and nitrogen. An average car needs only approximately 30 ml of NH₃ per 100 km to neutralize any NOx emissions. If the vehicles run with NH₃ as fuel, this amount is unimportant with respect to the fuel tank volume.

Ammonia is used as fertilizer in the agriculture sector. It is also converted into urea by reacting with CO₂. The majority of growth in ammonia usage is expected to be for industrial uses and the production of fertilizer products.

It is also worth it to examine the option to cool the engine with ammonia that can act as a refrigerant while it is heated to the temperature at which it is fed to the power producer (ICE or

fuel cell). Optionally, the cooling outcome of ammonia, i.e., its high latent heat of evaporation, may be used to harvest some air conditioning onboard.

IS AMMONIA REALLY A FUEL?

Ammonia as a sustainable fuel can be used in all types of combustion engines, gas turbines, and burners with only small modifications and directly in fuel cells. Ammonia was initially used as a fuel for buses in Belgium in 1940s [2]. Many studies have already been performed and many applications have been implemented so far. A prototype unit for combustion which enabled liquid kerosene and gaseous ammonia to be fed, and ammonia was combusted in a gas-turbine unit. Further studies have been performed by various researchers which have proven the practicality of using ammonia as fuel [3, 4, 5, 6, 7, 8, 9, 10]. Numerical studies of combustion characteristics of ammonia as a renewable fuel have been conducted. Ammonia can also be used as a fuel blending option for current gasoline and diesel engines. Combustion and emissions characteristics of compression-ignition engine using dual ammonia-diesel fuel have been performed. Performance enhancement of ammonia-fueled engine by using dissociation catalyst has been studied. These are just a few examples to show the current progress in the ammonia utilization options in transportation applications.

IS AMMONIA SUITABLE FOR TRANSPORTATION SECTOR?

The storage and delivery infrastructure of ammonia is similar to liquefied petroleum gas (LPG) process. Under medium pressures (5-15 bar), both of the substances are in liquid form which brings the significant advantage because of storage benefits. Today, vehicles running with propane are mostly accepted and used by the public since their on-board storage is possible and it is a good example for ammonia-fueled vehicle opportunities since the storage and risk characteristics of both substances are similar to each other. An ammonia pipeline from the Gulf of Mexico to Minnesota and with divisions to Ohio and Texas has served the ammonia industry for many years. It indicates that there is a working ammonia pipeline transportation infrastructure which can be spread worldwide. The potential of ammonia usage in many applications will be dependent on the availability of ammonia in urban areas. Ammonia is a suitable substance to be

transferred using steel pipelines with minor modifications which are currently used for natural gas and oil. In this way, the problem of availability of ammonia can be eliminated.

HOW CAN AMMONIA BE USED IN TRANSPORTATION?

Ammonia has significant potential as an alternative fuel to further the sustainable development of transportation sector. A few of the following alternatives are shown in Fig. 2 for direct ammonia usage in transportation applications.

Currently, the majority of the locomotive fleet is made up of diesel-electric locomotives, operating with either two-stroke or four-stroke prime mover diesel engines that is coupled to an electric generator. Application of ammonia fuel for internal combustion engine (ICE) with the alternative locomotive configurations direct feed, or a combination of direct feed and decomposition subcategory options will bring more sustainable solutions. Additionally, fuel cell driven vehicles and locomotives may contribute to solve the associated matters of urban air superiority and national energy security influencing the rail and transportation sector.

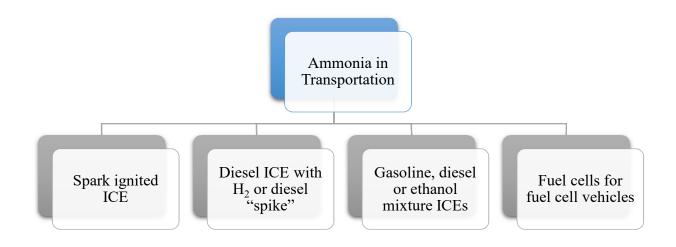


Fig. 2. Ammonia utilization options in transportation sector

IS AMMONIA A CLEAN FUEL?

Compared to gasoline vehicles, ammonia-fueled vehicles do not produce direct CO₂ emissions during operation. Since ammonia produces mainly water and nitrogen on combustion,

replacing a part of conventional fuel with ammonia will have a large effect in reducing carbon dioxide emissions.

WHAT FACTORS SHOULD BE CONSIDERED IN SETTING RFS TARGETS POST-2020?

- Environmental impact (including the indirect land-use changes)
- Cost
- Availability of clean production routes (e.g. solar, hydropower, wind)
- Blending properties (e.g. combustion characteristics)
- Fuel production life cycle emissions
- Fuel storage and transport infrastructure
- Integrability
- Multi-purpose usage (e.g. power, cooling, heating)

WHICH ENVIRONMENTAL IMPACTS SHOULD BE CONSIDERED?

A life cycle is the set of phases of a product or service system, from the extraction of natural resources to last removal. Overall environmental impact of any process is not complete if only operation is considered, all the life steps from resource extraction to disposal during the lifetime of a product or process should be considered. The selection of future vehicle options can strongly depend on the emission characteristics. As the world struggles with greenhouse gas emission reduction policies, global warming potential is the main characteristic to compare the total CO₂ equivalent emission from the alternative vehicles. Abiotic depletion, human toxicity, ozone layer depletion appear to play an important role for decision of using clean transportation vehicles because there are a vast amount of road vehicles in the cities which can cause severe side effects. Moreover, when considering alternative fuels, issues such as land use, fertilizer use, water for irrigation, waste products etc. are necessary points to be addressed. Therefore, indirect land-use changes (ILUC) should also be considered. Indirect land-use changes can also have important social and environmental impacts which can include biodiversity, water quality, food prices and supply, community and cultural stability. Assessing the indirect land-use changes is known as a

challenging topic. Some methods of quantifying indirect land-use changes can be listed as follows:

- Implementing empirical calculations based on previously experienced indirect land-use changes
- Developing life cycle analyses methodology with lower uncertainty ranges
- Developing integrated models combining the life cycle, sustainability, efficiency, social cost etc.

The following environmental impact categories represent higher significance in life cycle assessment approach, hence suggested to be included in decision making processes:

- Global warming potential is the main characteristics to be compare the total CO₂ equivalent emission from any source.
- Abiotic resources are natural resources including energy resources. Since fossil fuels resources are declining gradually, abiotic depletion potential is also a significant category.
- Human toxicity may play an important role for decision of using alternative fuels.
- Acidification potential is for acidifying substances which causes a wide range of impacts on soil, groundwater, surface water, organisms, ecosystems and materials.
- Marine aquatic eco-toxicity refers to impacts of toxic substances on marine aquatic ecosystems which is more important for maritime transportation sector.
- Land occupation/land use refers to the total arrangements, activities and inputs undertaken in
 a certain land cover type. The term land use is also used in the sense of the social and
 economic purposes for which land is managed.

HOW MUCH GREENHOUSE GAS CAN I SAVE IF I DRIVE AN AMMONIA CAR?

Considering a complete life cycle counting the production, transport and usage of the fuel, a diesel driven car can emit greenhouse gas emissions of about 220 g per km. An ammonia driven car can decrease this number down to about 70 g per km if it is produced from solar energy and about 150 g per km if it is produced from hydrocarbon cracking.

IS AMMONIA A COST EFFECTIVE FUEL?

The illustrative cost comparison of various fueled vehicles is shown in Fig. 3 and 4. Considering the current market prices of the fuels, ammonia is the lowest cost fuel corresponding to about 3.1 US\$ in a 100 km driving range. This shows that ammonia is a promising transportation fuel in terms of cost. There is an advantage of by-product refrigeration which reduces the costs and maintenance during vehicle operation. Some additional advantages of ammonia are commercial availability and viability, global distribution network and easy handling experience. Ammonia is a cost effective fuel per unit energy stored onboard compared to methanol, CNG, hydrogen, gasoline and LPG as shown in Fig. 3.

In Table 1, the exact fuel energy per mass is given regarding fuel's higher heating value. The volumetric energy of the fuel is found by multiplying the HHV with the density value listed in the third column. Ammonia's HHV is around half of the one of gasoline, and its density is also inferior. Therefore liquid NH₃ stores 2.5 fewer energy per unit capacity than gasoline. If the NH₃ is stored in the form of hexaamminemagnesium chloride to remove the hazard related to its toxicity, the energetic cost to pay for discharging ammonia reduces its HHV. Among alternative fuels, ammonia yields the lowest cost per energy basis. Therefore, it is important to note that low-carbon transportation projects should be eligible for compliance purposes.

Table 1. Comparison of ammonia with other fuels

Fuel/storage	Pressure (bar)	Density (kg m ⁻³)	HHV (MJ kg ⁻¹)	HHV per Volume (GJ m ⁻¹)	Energy per Volume (GJ m ⁻¹)	Cost per Mass (US\$ kg ⁻¹)	Cost Per Volume (US\$ m ⁻¹)	Cost per Energy (US\$ GJ ⁻¹)
Gasoline, C ₈ H ₁₈ /liquid	1	736	46.7	34.4	34.4	1.03	754.86	21.97
CNG, CH ₄ /integrated storage	250	188	42.5	10.4	7.8	0.91	170.60	21.29
LPG, C ₃ H ₈ /pressurized tank	14	388	48.9	19	11.7	1.06	413.66	21.74
Methanol, CH ₃ OH/liquid	1	786	14.3	11.2	9.6	0.41	317.80	28.31
Hydrogen, H ₂ /metal hydrides	14	25	142	3.6	3	3.02	75.49	21.29

Ammonia, NH ₃ /pressurized tank	10	603	22.5	13.6	11.9	0.23	136.63	10.04
Ammonia, NH ₃ /metal amines	1	610	17.1	10.4	8.5	0.23	138.14	13.21

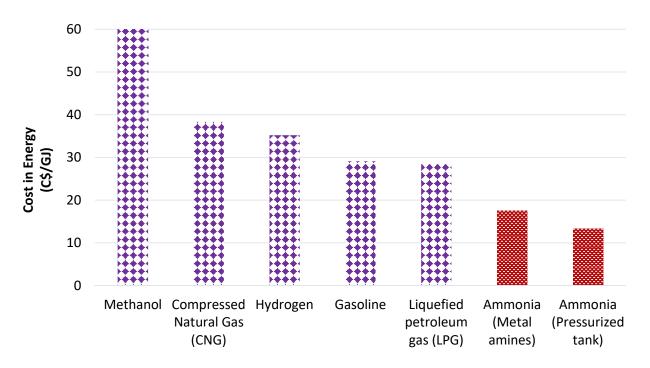


Fig. 3. Comparison of various vehicle fuels in terms of energy cost per gigajoule

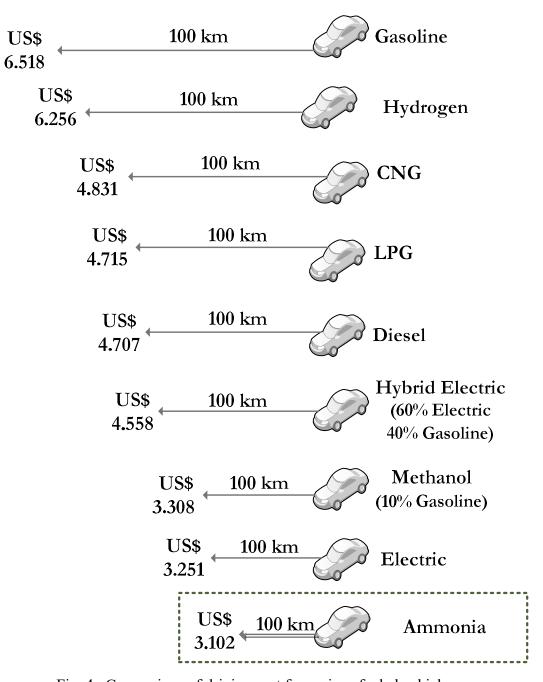


Fig. 4. Comparison of driving cost for various fueled vehicles

WHAT IS THE PROCESS OF AMMONIA PRODUCTION?

The most common ammonia synthesis technique is recognized as Haber-Bosch process in the world. In this process, nitrogen is supplied through air separation process. Hydrogen is mainly supplied using steam methane reforming or coal gasification. However the source of hydrogen can be renewable resources. The Haber-Bosch is an exothermic process that combines hydrogen and nitrogen in 3:1 ratio to produce ammonia. The reaction is facilitated by catalyst and the optimal temperature range is 450-600°C.

Alternative production pathways are also available and under investigation including electrochemical and biological routes. These routes can easily be integrated to renewable energy sources for cleaner production. The electrochemical process can be carried out under ambient conditions or at higher temperatures depending on the type of the electrolyte material used. There are various electrochemical pathways such as molten salt, polymer membrane, liquid electrolyte etc. intensively being researched at the moment [11, 12, 13, 14, 15, 16].

The electrochemical process can be carried out under ambient conditions or at higher temperatures depending on the type of the electrolyte material used. For high temperature electrolytic routes of ammonia production, the use of waste heat from thermal or nuclear power plants or heat from renewable energy sources like solar would make the overall process more environmentally friendly.

Ammonia's key advantages is its ability to be used as a storage medium. Renewable energy generation does not often match electrical demand which causes a requirement of storage. Green ammonia can be manufactured from surplus renewable sources, which would reduce the amount of electricity exported to neighbouring jurisdictions at a negative cost.

WHAT IS THE SOURCE OF AMMONIA AND IS IT CLEANER THAN OTHER FUELS?

In terms of conventional resources, naphtha, heavy fuel oil, coal, natural gas coke oven gas and refinery gas can be used as feedstock in ammonia production. Natural gas is the primary feedstock used for producing ammonia worldwide corresponding to about 72%. However, renewable resources can easily be integrated for ammonia production. In this way, decentralized ammonia production can be realized which further decreases the delivery cost of the fuel. Many studies have been performed to investigate the ammonia production routes and their environmental impacts. Here, some of them are briefly shown.

The production of the different fuels is compared in terms of abiotic depletion of sources as shown in Fig. 5. Ammonia fuel has the lowest abiotic depletion value compared to others although the production process may be fossil fuel based. There are multiple pathways for ammonia production. Ammonia is cleaner when produced from renewable resources. Fig. 6

shows the comparison of ozone layer depletion values for various transportation fuels. Ammonia has lowest ozone layer depletion even if it is produced from steam methane reforming and partial oxidation of heavy oil.

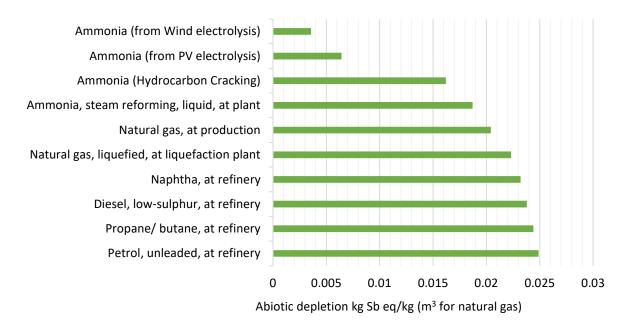


Fig. 5. Abiotic depletion values during production of various fuels

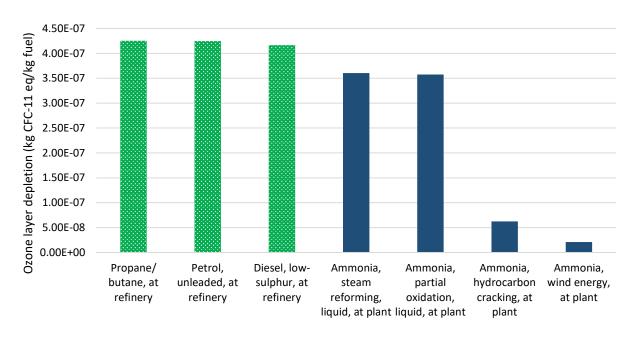


Fig. 6. Ozone layer depletion during productions of various fuels

Fig. 7 compares the total greenhouse gas emissions during production of 1 MJ energy from various resources including gasoline, LPG, diesel, natural gas and ammonia. Production of 1 MJ energy from ammonia has lower emissions than gasoline, LPG, diesel, oil and natural gas.

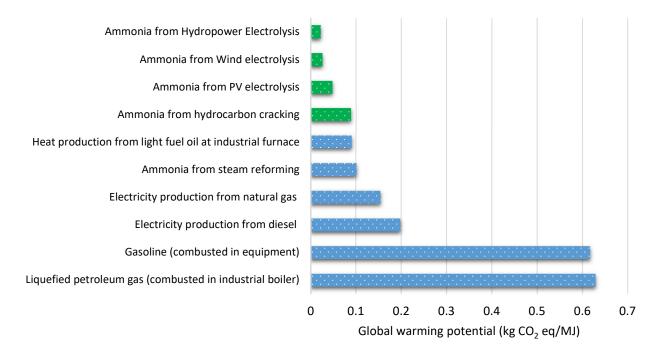


Fig. 7. Comparison of global warming potential of 1 MJ energy production from various resources

Giving priority for complete conversion from fossil fuel based fuels to carbon-free fuels will bring short term and long term solutions to combat global warming. Therefore, the activities to lower the carbon intensity of conventional transportation fuels be should eligible for compliance purposes.

AMMONIA IN ROAD TRANSPORTATION

Assessing the life cycle emissions from vehicles is a powerful method for transportation sector. The usage of fossil fuel based electricity decreases the attractiveness of electric (EV) and hybrid electric (HEV) vehicles. Henceforth, noteworthy attention should be paid to the power generation technologies and their CO₂ intensity, used to supply electricity to EVs or HEVs. The GHG emissions of EVs and HEVs are therefore dependent on the CO₂ intensity of the energy

mix and differs based on the countries. A characteristic life cycle of a vehicle technology can be categorized into two main steps, namely fuel cycle and vehicle cycle. In the fuel cycle, the processes beginning from the feedstock production to fuel utilization in the vehicle are considered. In the vehicle cycle, utilization of fuel is considered. Among the selected categories, global warming, abiotic depletion and human toxicity results carry more significant decision parameters for road vehicles. The results presented here are given on per km basis based on the fuel consumption rates given in Table 2.

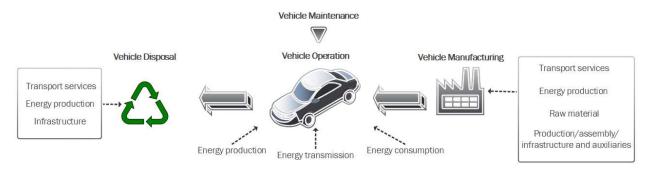


Fig. 8. Complete life cycle of vehicles including fuel/vehicle cycle.

Table 2. Energy consumptions per km for the selected vehicles

Fuel		Fuel/Energy Consumption	Unit
Gasoline		0.0649108	kg/km
Diesel		0.0551536	kg/km
M90	Methanol	0.1180535	kg/km
14150	Gasoline	0.0060664	kg/km
Hydrogen		0.0195508	kg/km
Ammonia		0.0926600	kg/km
EV		0.2167432	kWh/km
HEV	Electric	0.1083716	kWh/km
TIL V	Gasoline	0.0324554	kg/km
CNG		0.0603914	kg/km
LPG		0.057629687	kg/km

The specific conditions for the selected vehicles are presented herein:

- Gasoline: All processes on the refinery site excluding the emissions from combustion facilities, including waste water treatment, process emissions and direct discharges to rivers are accounted for. The inventory data also includes the distribution of petroleum product to the final consumer including all necessary transports. Transportation of product from the refinery to the end user is considered together with operation of storage tanks and petrol stations. Emissions from evaporation and treatment of effluents are accounted for. Particulate emissions cover exhaust- and abrasions emissions.
- Diesel: Diesel is evaluated as low-Sulphur at regional storage with an estimation for the total conversion of refinery production to low-Sulphur diesel. An additional energy use (6% of energy use for diesel production in the refinery) has been estimated. The other processes are similar to gasoline. Particulate emissions cover exhaust- and abrasions emissions.
- CNG: Natural gas with a production mix at service station is taken into account. The inventory data contains electricity necessities of a natural gas service station together with emissions from losses. The data set represents service stations with high (92%), medium (6%) and low (2%) initial pressure. VOC emissions are obtained from gas losses and contents of natural gas. Particulate emissions cover exhaust- and abrasions emissions.
- Hydrogen: Hydrogen is produced during cracking of hydrocarbons. It includes combined data for all processes from raw material extraction until delivery at plant. The output fractions from an oil refinery are composite combinations of mainly unreactive saturated hydrocarbons. The first processing step in converting such elements into feedstock suitable for the petrochemical industries is cracking. Essentially a cracker achieves two tasks in (i) rising the complexity of the feed mixture into a smaller number of low molecular mass hydrocarbons and (ii) presenting unsaturation into the hydrocarbons to enable more reactivity. The raw hydrocarbon input from the refinery is fed to the heater unit where the temperature is increased. The forming reaction products vary based on the composition of the input, the temperature of the heater and the residence time. The cracker operator selects temperature and residence time to enhance product mix from a supplied input. Cracker feeds can be naphtha from oil refining or natural gas or a mixture of both. After exiting the heater, the hydrocarbon gas is cooled to prevent extra reactions. After that, it is sent to the separation phase where the individual hydrocarbons are separated from one another by fractional distillation. Particulate emissions

cover exhaust- and abrasions emissions. In order to have comparable results where hydrogen comes from non-fossil fuels such as solar PV and nuclear, they are also taken into account in the analyses by applying water electrolysis route. The electrolyzer is assumed to consume about 53 kWh electricity for one kg of hydrogen production.

- Ammonia: Ammonia synthesis process is Haber-Bosch which is the most common method in the world. Ammonia production requires nitrogen and hydrogen. In this study, hydrogen is assumed to be from hydrocarbon cracking as explained in the previous paragraph. Cryogenic air separation is mostly used method for massive amount of nitrogen production. In the life cycle assessment of nitrogen production, electricity for process, cooling water, waste heat and infrastructure for air separation plant are included. Haber-Bosch process is an exothermic method that combines hydrogen and nitrogen in 3:1 ratio to produce ammonia. The reaction is facilitated by catalyst (iron-oxide based) and the optimal temperature range is 450-600°C. Particulate emissions cover exhaust- and abrasions emissions. In order to have comparable results where ammonia comes from non-fossil fuels such as solar PV and nuclear, they are also taken into account in the analyses. The generated hydrogen from electrolyzers are used for ammonia synthesis plant.
- EV: Electricity consumption is included. Particulate emissions comprise exhaust and abrasions emissions. Heavy metal emissions to soil and water caused by tire abrasion are accounted for. In the electricity usage process, electricity production mix, the transmission network and direct SF6-emissions to air are included. In order to present a renewable based scenario for electric vehicles, a mixture of renewables for energy requirement during the operation are also evaluated consisting of 25% biomass, 25% solar PV, 25% wind power and 25% hydropower.
- HEV: Hybrid car is assumed to be 50% electric and 50% gasoline with ICE. Electricity and gasoline consumptions are included. Particulate emissions comprise exhaust and abrasions emissions. Heavy metal emissions to soil and water caused by tire abrasion are accounted for. For the hybrid vehicle's electricity, a mixture of renewables for energy requirement during the operation are also evaluated consisting of 25% biomass, 25% solar PV, 25% wind power and 25% hydropower.
- Methanol: The selected fuel M90 consists of 90% methanol and 10% gasoline. The raw materials, processing energy, estimate on catalyst use, and emissions to air and water from

process, plant infrastructure are included. The process describes the production of methanol from natural gas via steam reforming process to obtain syngas for the production of methanol. There is no CO₂ use and hydrogen is assumed as burned in the furnace. Raw materials, average transportation, emissions to air from tank storage, estimation for storage infrastructure are included for the distribution part where 40% of the methanol is assumed to be transported from overseas. Particulate emissions cover exhaust- and abrasions emissions.

• LPG: All processes on the refinery site excluding the emissions from combustion facilities, including waste water treatment, process emissions and direct discharges to rivers are considered. All flows of materials and energy due to the throughput of 1 kg crude oil in the refinery is accounted for. Refinery data include desalting, distillation (vacuum and atmospheric), and hydro treating operations. Particulate emissions cover exhaust- and abrasions emissions.

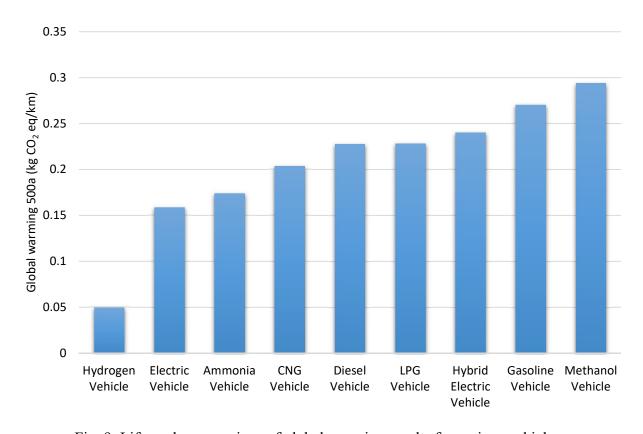


Fig. 9. Life cycle comparison of global warming results for various vehicles

The global warming potentials of assessed vehicles are comparatively shown in Fig. 9. The lowest GHG emissions are observed in hydrogen, electric and ammonia vehicles corresponding to 0.049 kg CO₂ eq/km, 0.15 kg CO₂ eq/km and 0.17 kg CO₂ eq/km, respectively. Hydrogen consumption is quite lower than ammonia consumption in the passenger car because of higher energy density. It is an expectable result that EVs also yield lower global warming potential, however production pathway of electricity has a key role in GHG emissions. If electricity production can be realized by renewable sources such as solar, biomass, hydropower and wind energy, total emissions would decrease for both EVs and HEVs.

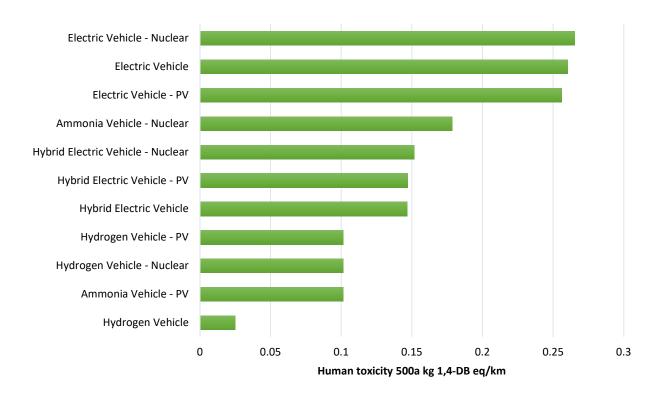


Fig. 10. Life cycle comparison of human toxicity results for various vehicles from nuclear energy and solar PV routes

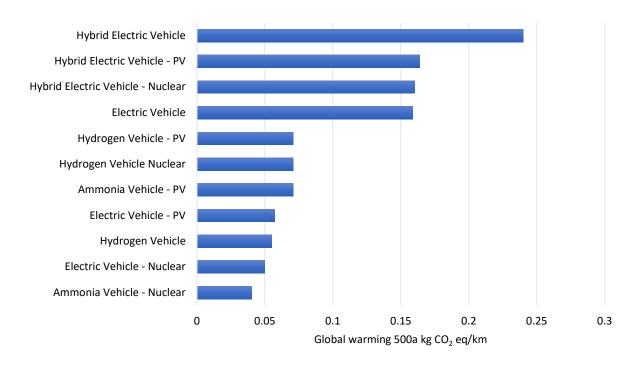


Fig. 11. Life cycle comparison of global warming results for various vehicles from nuclear energy and solar PV routes

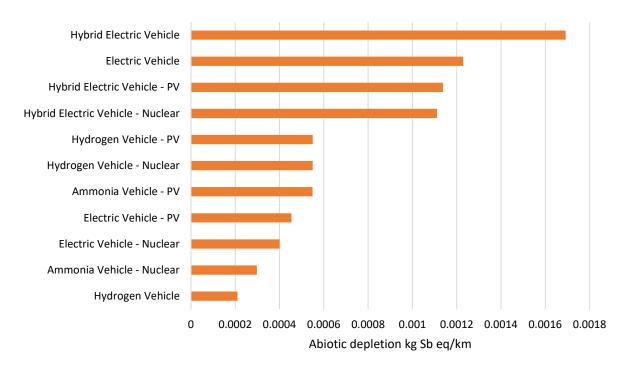


Fig. 12. Life cycle comparison abiotic depletion for various vehicles from nuclear energy and solar PV routes

The electricity from nuclear (25%), biomass (25%), hydropower (25%) and solar PV (25%) are equally used only for the operation processes of the EVs and HEVs vehicles as renewable mix. Utilization of renewable electricity for EVs, HEVs, ammonia and hydrogen vehicles are comparatively shown for various impact categories. For human toxicity category, using solar and nuclear energy does not cause significant reductions for EVs and HEVs as shown in Fig. 10. However, for global warming and abiotic depletion categories, both solar energy and nuclear energy routes lower the environmental impacts more than 50%. Ammonia driven vehicle where the ammonia is produced from nuclear electrolysis method yield the lowest GHG emissions corresponding to about 0.04 kg CO₂ eq. per km as shown in Fig. 11. Hydrogen and EVs (from nuclear and PV) have quite similar greenhouse gas emissions in the range of 0.049-0.057 kg CO₂ eq. per km. In terms of abiotic depletion values, hydrogen vehicle still yields the lowest value whereas nuclear routes for ammonia and EVs further decrease the abiotic depletion impact as shown in Fig. 12.

The results show that hydrogen and ammonia vehicles are the most environmentally benign ones in most of the environmental impact categories. Ammonia as a sustainable and clean fuel has lowest global warming potential after EVs and yield lower ozone layer depletion values than EVs. However, in case renewables are used both for ammonia vehicles and EVs, ammonia can suggest lower environmental impacts. Although EVs do not emit direct CO₂ during operation, the production and disposal processes of batteries bring some consequences which harm the environment in terms of acidification, eutrophication and human toxicity.

AMMONIA IN MARITIME TRANSPORTATION

Decreasing the global warming potential caused by current transportation technologies and fuels can be reduced significantly by replacing alternative clean fuels. Sea transportation vehicles mostly use heavy fuel oil or diesel fuel for power generation. Ocean tankers and freight ships require massive amount of energy for operation.

Being a sustainable energy carrier that can be generated from any primary energy source, ammonia can subsidize to a broadening of maritime fuel resources and may offer the long term option of being generated from renewable resources.

A brief life cycle consideration of maritime vehicles, ocean tanker and freight ship, is shown here where the tankers are driven with hydrogen and ammonia instead of heavy fuel oils in the power engines. Additionally, dual fuel options, heavy fuel oil and hydrogen/ammonia, are investigated.

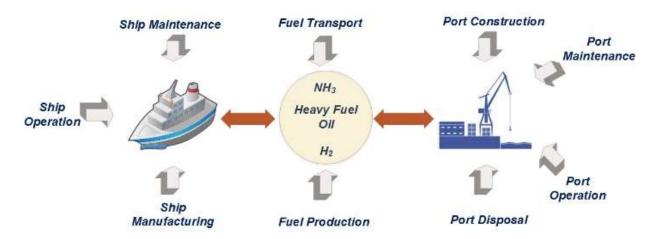


Fig. 13. Life cycle steps of maritime transportation

A comparative life cycle assessment of a transoceanic freight ship and transoceanic tanker, is performed to examine the effects of clean fuel driven maritime vehicles on the environment. The complete transport life cycle is evaluated in the life cycle analyses comprising of manufacture of tanker/freight ship; operation of tanker/freight ship; construction and land use of seaport; operation, maintenance and disposal of seaport; production and transportation of hydrogen and ammonia. A tonne kilometer by shipping is defined as unit of measure of goods transport which represent the transport of one tonne by a vessel over one kilometer.

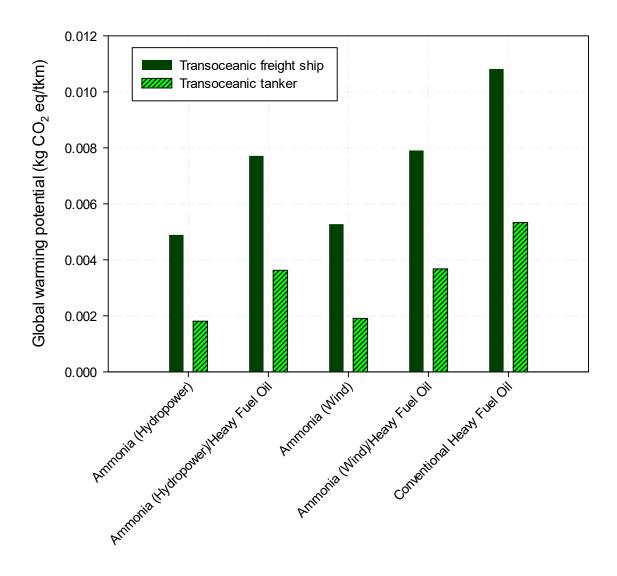


Fig. 14. Global warming potential of transoceanic tanker and transoceanic freight ship per tonne kilometer for clean fuels and conventional heavy fuel oil

The heavy fuel oil combustion significantly contributes to many of the environmental impact. Using ammonia as dual fuel in the marine engines can decrease total greenhouse gas emissions up to 34.5% per tkm. Similarly, sole ammonia driven transoceanic tanker releases about 0.0018 kg CO₂ eq/tkm greenhouse gas compared to 0.0055 kg CO₂ eq/tkm for sole heavy fuel oil tanker as shown in Fig. 14.

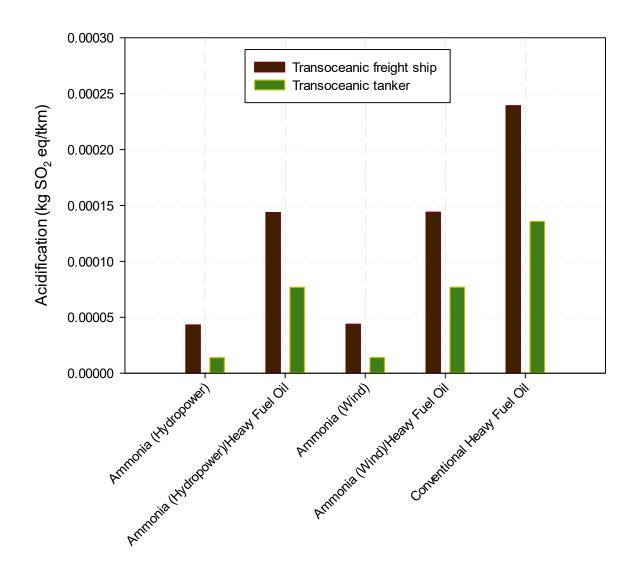


Fig. 15. Acidification values of transoceanic tanker and transoceanic freight ship per tonne kilometer for clean fuels and conventional heavy fuel oil

The acidification values of heavy fuel oil driven transoceanic tanker and freight ship are mainly caused by SO₂ and NO_x emissions (Fig. 15) which corresponds to more than 90% of overall acidification value. The source of SO₂ emission is predominantly the operation of tanker and freight ship (96.8%). This is caused by the sulfur content of the heavy fuel oil hence it is mostly eliminated if ammonia is used.

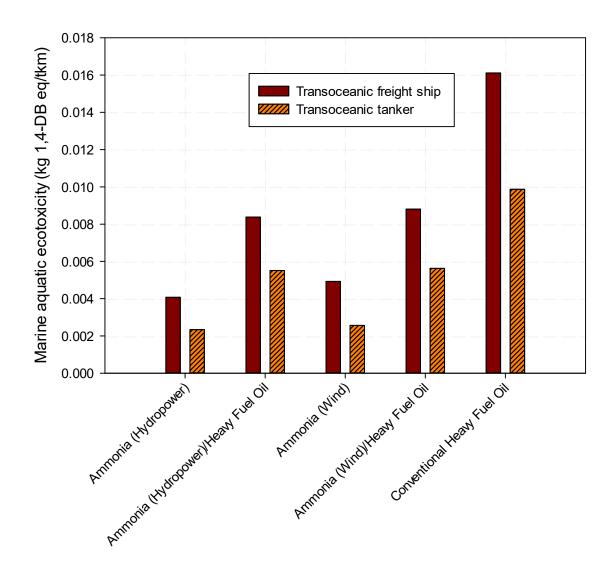


Fig. 16. Marine aquatic ecotoxicity values of transoceanic tanker and transoceanic freight ship per tonne kilometer for clean fuels and conventional heavy fuel oil

Using ammonia (wind energy) in a transoceanic tanker as dual fuel with heavy fuel oil lowers the ecotoxicity level about 47% as shown in Fig. 16. For conventional heavy fuel oil driven transoceanic tanker, crude oil is the main depleted source similar to dual fuel options as shown in Fig. 17. For ammonia, the depletion is mainly caused by port operation process rather than tanker/ship operation. As non-carbon clean fuels for maritime ship engines, ammonia yields considerably lower global warming impact during operation. This demonstrates that if clean

fuels are even partially replaced with current hydrocarbon derived fuels, total GHG emissions in maritime transportation can be lowered significantly.

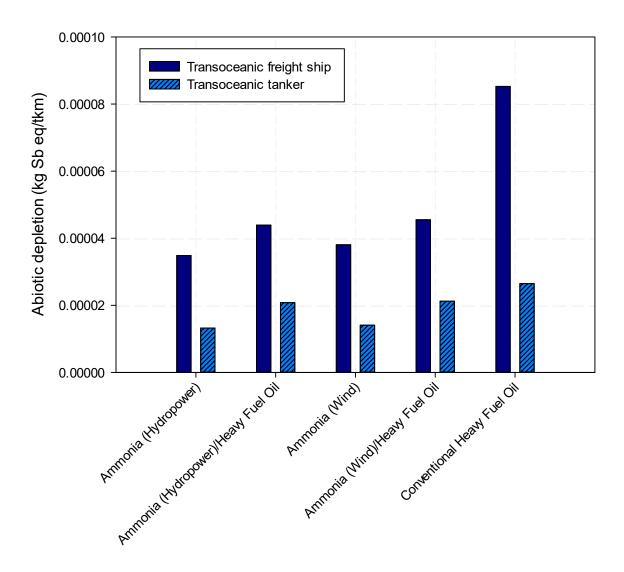


Fig. 17. Abiotic depletion values of transoceanic tanker and transoceanic freight ship per tonne kilometer for clean fuels and conventional heavy fuel oil

AMMONIA IN AVIATION

Petroleum based fuels have the combination of accessibility, ease of handling, energy content, performance, and price because of being a mature product. Therefore, these type of fuels are heavily used by the transportation sector including air, road, and sea. However, limited

nature, nonhomogeneous source distribution, changing prices, and end use related emissions of fossil fuels have driven most of the industries such as air transportation to search for other alternatives. Aviation is responsible for 12% of CO₂ emissions from all transports sources. The national and international flights overall the world produced 770 million tons of CO₂ in 2015 [17]

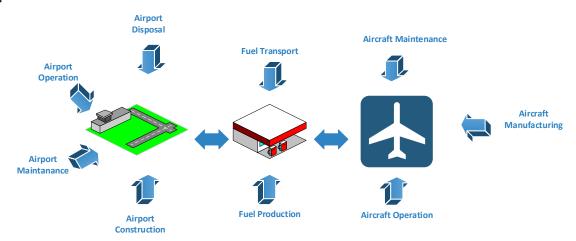


Fig. 18. Illustration of complete well-to-wake life cycle phases used in the LCA study

The overall life cycle emissions of an aircraft running on various aviation fuels are calculated from well-to-wake. The life cycle phases included in the analyses (shown in Fig. 18) are as follows: (i) production, operation and maintenance of the aircraft, (ii) construction, maintenance and disposal of the airport, (iii) production, transportation and utilization of the aviation fuel in the aircraft.

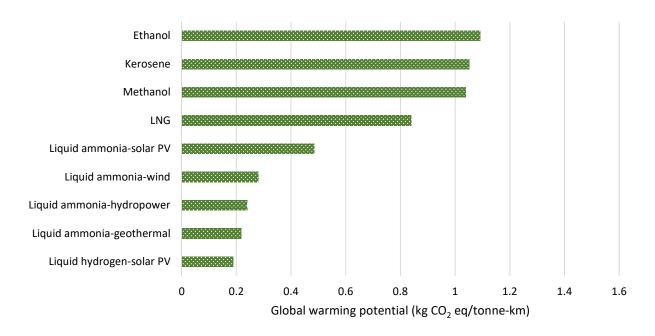


Fig. 19. Global warming potential of various fueled aircrafts per travelled tonne-km

The overall GWP value of ethanol is the highest corresponding to 1.09 kg CO₂ eq/tonne-km since it comes from ethylene which is a hydrocarbon as shown in Fig. 19. Renewable energy usage in the fuel production processes decreases the overall GHG emissions. The hydropower based ammonia fueled aircraft releases about 0.24 kg CO₂ eq per tonne-km. This value goes down to 0.21 for kg CO₂ eq/tonne-km for ammonia fueled aircraft and 0.18 kg CO₂ eq/tonne-km for hydrogen fueled aircraft in case the renewable source is solar energy. Although there are several processes, operation of the aircraft is the second largest contributor corresponding to 34%. Operation and maintenance of the airport is the primarily responsible for GWP corresponding to 48.9% in total where it is distributed into sub-processes such as natural gas burning in the furnace (22%), light fuel oil burning in the furnace (5%), lignite burning in the power plant (7%) and hard coal burning (8%)

Renewable based ammonia driven aircrafts can significantly decrease the depletion of abiotic resources down to 0.0014 kg Sb eq/tonne-km which corresponds to about 10% of conventional steam methane reforming based ammonia fueled aircraft as shown in Fig. 20.

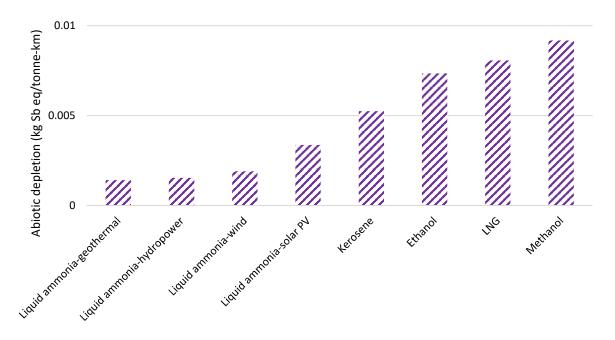


Fig. 20. Abiotic depletion values of various fueled aircrafts per travelled tonne-km

It is common knowledge that of HC, CO, NO_x, PM, and CO₂ which are named as GHG emissions cause environmental damages and adverse effects on human health. The marginal external price of a unit of these emissions is identified as environmental and social costs of emissions. Environmental and social costs of HC, CO, NO_x, PM, and CO₂ emissions of various fueled aircrafts are evaluated in terms of USD/tonne-km as shown in Fig. 21. renewable based ammonia yields lower environmental and social cost of emissions in comparison with kerosene.

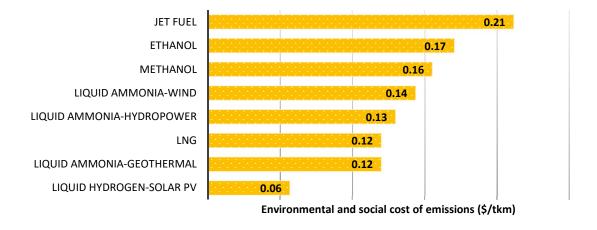


Fig. 21. Total environmental and social cost of emissions for various fueled aircrafts from conventional and renewable resources

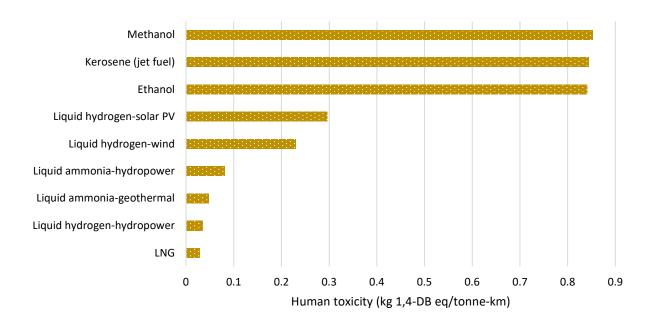


Fig. 22. Human toxicity potential of various fueled aircrafts per travelled tonne-km

Among all alternative aviation fuels, the kerosene jet fuel, methanol and ethanol human toxicity values are greater. Ammonia fuel from hydropower resource has comparable toxicity value corresponding to 0.08 kg 1,4- DB eq/tonne-km with LNG corresponding to 0.03 kg 1,4- DB eq/tonne-km as shown in Fig. 22.

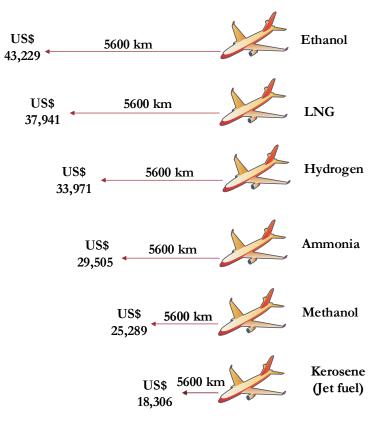


Fig. 23. Comparison of fuel costs during the operation of aircrafts for the given range

Table 3. Average fuel consumption rates and fuel costs for selected alternative fuels

Fuel	Fuel Consumption (kg/km)	Fuel Consumption (kg/tonne-km)	Fuel Cost (USD/kg)	
Kerosene	7.99	0.217666015	0.409	
(Jet fuel)	7.99	0.217666815	0.409	
Methanol	18.06	0.492185866	0.250	
Ammonia	18.82	0.512730362	0.280	
Hydrogen	2.64	0.071866485	2.300	
LNG	9.46	0.257732767	0.716	
Ethanol	12.47	0.339891348	0.619	

Fig. 23 illustrates the cost of flight for a 5600 km distance in which alternative fuels are used for the aircrafts when the fuel costs and consumption rates in Table 3 are considered.

Although LNG represented better environmental performance, the cost of aircraft operation in terms of fuel is the highest for ethanol and LNG. In the calculations, current conventional based routes are taken into account for comparison purposes. Hydrogen fueled aircraft has lower cost compared to these two alternatives. Since the production of kerosene from crude oil is a more mature technology, it represents the lowest cost among all. Liquid ammonia is also a low cost alternative compared to hydrogen.

Transparency in Renewable Fuels

Ontario's Cap & Trade Program states:

The Renewable Fuel Standard is a technology-neutral policy and a technology-neutral approach lets the alternatives compete on their merits.

In order for renewable fuel technology to compete on its own merits and for the market to decide which technology has the best merits, energy consumers must ultimately participate in selection of renewable fuels.

Ontario's Cap & Trade program has the following participants:

Mandatory	Facilities and natural gas distributors with emissions of 25,000 tonnes or more of
participants	greenhouse gas emissions per year are required by law to participate in the
	program. Additionally, fuel suppliers that sell more than 200 litres of fuel per
	year and electricity importers must also participate in the program.
Voluntary	Facilities generating more than 10,000 tonnes but less than 25,000 tonnes of
participants	emissions may choose to opt in to the program. These companies will be subject
	to the same rules as mandatory participants.
Market	Companies that do not have emissions to report and therefore do not have a
participants	compliance obligation can also participate in the auction. Market participants
	can include individuals, not-for-profit organizations and companies without
	compliance obligations.

For the market to decide the winners in the Cap & Trade program, each energy consumer must also be a Cap & Trade participant.

This could be administered through the income tax system, with each family or business entity being issued its own baseline carbon credit. At income tax time, each energy or fuel supplier would provide each consumer a carbon statement that the energy consumer would

include in their income tax return. Depending upon each tax entity's carbon consumption, the net carbon credits would translate into additional carbon tax payments or refunds.

TRANSPARENCY VIA A FLOW-THROUGH CARBON TAX

From an energy consumer's perspective, it is impractical to determine one's own individual emissions. However, it is simple to determine their expected emissions by the amount of fuel consumed. Therefore, a more practical and transparent way of putting a cost on carbon emissions would be through a flow-through carbon taxation program similar to the GST/HST system. The carbon tax would replace the fuel excise tax and would be applicable to ALL consumable energy and fuel.

This taxation system would start at the producer level. For energy produced in Ontario, the carbon tax would replace the fuel excise tax. The carbon tax would be based on the carbon content of the fuel sourced from fossil sources or of the amount of fossil-carbon consumed to convert a fuel to a consumable form of energy. The carbon tax would then flow through from the producer to the distributor to the consumer.

For a technology-neutral carbon policy, there must be a completely level playing field and there is more to the life cycle environmental impact of the fuel than the amount of carbon dioxide generated from the its combustion. There are nine categories of life cycle environmental impacts caused by the production and utilization of energy that must be considered for a completely level playing field:

Environmental	Description
Impact	
Abiotic Depletion	Abiotic resources are natural resources including energy resources.
	Since fossil fuels resources are declining gradually, abiotic depletion
	potential is also a significant category. It is expressed in kg antimony
	(Sb) equivalents.

Description
Acidification potential is for acidifying substances which causes a wide
range of impacts on soil, groundwater, surface water, organisms,
ecosystems and materials. kg SO2 equivalents is used to express the
acidification potential.
Eutrophication is the impact of excessive levels of macro-nutrients in
the environment which is mainly caused by disposal processes. It is
stated as kg PO ₄ equivalents.
Global warming potential is the main characteristics to be compare the
total CO2 equivalent emission from any source. The greenhouse gasses
to air are related to the climate change. Methane, carbon monoxide etc.
are in this category.
Human toxicity may play an important role for decision of using
alternative fuels. It indicates the toxicity level and presented in 1,4-
dichlorobenzene (C ₆ H ₄ Cl ₂) equivalent.
Ozone layer depletion increases the amount of UVB that reaches the
Earth's surface. Laboratory and epidemiological studies demonstrate
that UVB causes non-melanoma skin cancer and plays a major role in
malignant melanoma development. Ozone depletion potential of
several gasses is specified in kg CFC-11 equivalent.
Terrestrial ecotoxicity refers to the potential for biological, chemical or
physical stressors to affect terrestrial ecosystems. Such stressors might
occur in the natural environment at densities, concentrations or levels
high enough to disrupt the natural biochemistry, physiology, behavior
and interactions of the living organisms that comprise terrestrial
ecosystems. The results are stated as 1,4-dichlorobenzene equivalents
per kg emission.

Environmental	Description
Impact	
Marine Aquatic Eco-	Marine aquatic eco-toxicity refers to impacts of toxic substances on
Toxicity	marine aquatic ecosystems which is more important for maritime
	transportation sector. It considers each substance emitted to the air,
	water or/and soil. The unit of this factor is kg of 1,4-dichlorobenzene
	equivalents (1,4-DB eq.) per kg of emission.
Land	Land occupation/land use refers to the total arrangements, activities
Occupation/Land	and inputs undertaken in a certain land cover type. The term land use is
Use	also used in the sense of the social and economic purposes for which
	land is managed. It is mainly expressed as m2a which implies area
	usage in a year.

Focusing on carbon dioxide alone precludes any consideration of the global warming potential of other greenhouse gas emissions. From the US EPA [18]:

Greenhouse Gas	Global Warming	Notes
	Potential	
	Over 100 years	
CO2	1	Regardless of the time period used, CO2 is
	by definition	the reference gas.
		CO2 remains in the climate system for a
		very long time: CO2 emissions cause
		increases in atmospheric concentrations of
		CO2 that will last thousands of years.

Greenhouse Gas	Global Warming	Notes
	Potential	
	Over 100 years	
Methane (CH4)	28–36	CH4 emitted today lasts about a decade on
		average, which is much less time than
		CO2. But CH4 also absorbs much more
		energy than CO2. The net effect of the
		shorter lifetime and higher energy
		absorption is reflected in the GWP. The
		CH4 GWP also accounts for some indirect
		effects, such as the fact that CH4 is a
		precursor to ozone, and ozone is itself a
		GHG.
		(Learn why EPA's U.S. Inventory of
		Greenhouse Gas Emissions and Sinks uses
		a different value.)
Nitrous Oxide (N2O)	265–298	N2O emitted today remains in the
		atmosphere for more than 100 years, on
		average.
Chlorofluorocarbons	Thousands	Sometimes called high-GWP gases
(CFCs),	or	because, for a given amount of mass, they
Hydrofluorocarbons	Tens of Thousands	trap substantially more heat than CO2.
(HFCs),		
Hydrochlorofluorocarbons		
(HCFCs),		
Perfluorocarbons (PFCs),		
Sulfur Hexafluoride (SF6)		

The carbon tax would be designed to improve upon the reduction of greenhouse gas emissions by including other detrimental effects on the environment. Rather than basing the tax

rate solely on the carbon content of the fuel, global warming would be one of the nine environmental impact categories.

Conceivably, some fuels could have a positive effect on a particular type of pollution. In this case, such a process should be given a negative environmental impact factor to reduce its tax rate. An environmental impact multiplier would then be applied to the baseline rate that takes into account the nine environmental impact categories. Each environmental impact category would have an assessment scale of -10 to 10. The environmental impact multiplier would then be based on the sum of the nine assessments.

Ultimately, only electricity or fuels with a zero or negative environmental impact assessment would have a zero carbon tax rate. The cost of the inputs of varying environmental impacts used to generate the each source of electricity would be included in the cost of electricity. This would ensure that that the lowest-cost electricity is produced with the lowest environmental impact.

MANUFACTURING PRODUCTS FROM FUELS

If part of the fuel was then converted to a product, the manufacturer of that carbon-containing product would then get a credit for the carbon that has been prevented from entering the atmosphere.

For example, some biofuel processes can take carbon out of the atmosphere and can permanently sequester it into the ground as biochar. This type of permanent sequestration can result in Terra Preta [19], which has enhanced soil fertility. This process would get a credit for the carbon content of the biochar, less whatever biofuel was consumed to produce the biochar.

For another example, if a manufacturer had a process to produce anhydrous ammonia and elemental carbon from natural gas, the manufacturer would pay the full carbon tax on the natural gas used in its process. It would then get a credit for the elemental carbon, less whatever natural gas was used for process heat. The elemental carbon would carry the maximum carbon tax but the consumer of the elemental carbon would get the full carbon credit if that elemental carbon

would be used to manufacture a carbon product, such as graphene. Since anhydrous ammonia has its own environmental impact, its carbon tax would be different than that of the natural gas.

TAX ADMINISTRATION

A carbon tax program would be easier to administer at a consumer level than a Cap & Trade program. Businesses are already familiar with the GST/HST system and would not require significantly more bureaucracy to administer. This would also be transparent because the carbon tax portion of the energy cost could be separately broken out in the price as well as the methodology used to calculate the pollution multiplier factors. Consumers would therefore be able to determine the most effective way of reducing their energy costs.

In addition, having a carbon tax would also create growth in the economy with businesses providing ways of reducing energy costs through new equipment sales or retrofitting existing equipment.

This carbon tax should be revenue-neutral and overseen by the Environmental Commissioner of Ontario. The entire carbon tax revenue stream should be collected in its own fund and not go into the general revenue fund. The carbon tax revenue should be returned to the citizens of Ontario in the form of an annual dividend at income tax time, less the cost of administering the program.

Since the carbon tax would replace the fuel excise tax, the government may take some money from the carbon tax fund equivalent to the tax that would have been collected from the fuel excise tax. This is necessary to keep the carbon-tax revenue-neutral. The equivalent revenue removed from the carbon tax fund should be used to fund public transit and/or climate change programs including research and innovation.

CARBON TAX VS CAP & TRADE

The website *carbontax.org* makes an excellent case for why a carbon tax is superior in terms of transparency to cap & trade for reducing carbon emissions:

CAP-AND-TRADE'S INHERENT DEFECTS [20]

A tax on carbon emissions isn't the only way to "put a price on carbon" and provide incentives to reduce use of high-carbon fuels. A **carbon cap-and-trade system** is an alternative approach supported by some prominent politicians, corporations and mainstream environmental groups. Cap-and-trade was the structure embodied in the Waxman-Markey climate bill that passed the House in 2009 but died in the Senate. And cap-and-trade is the cornerstone of the European Union's "Emissions Trading Scheme" (ETS).

Cap-and-trade systems can be effective under certain conditions. The U.S. *sulfur dioxide* cap-and-trade system instituted in the early 1990s efficiently reduced acid rain emissions from power plants. However, the scale of a *carbon* trading system — it would be up to 100 times larger than that for sulfur — combined with the lack of "technical fixes" for filtering or capturing CO2, rules out sulfur cap-and-trade as a model for carbon. Moreover, evidence from the EU's ETS suggests that <u>price volatility</u> and <u>gaming</u> by market participants have undermined the effectiveness of this complex, opaque indirect method of pricing carbon pollution.

The Carbon Tax Center along with <u>most economists</u> regard a carbon tax as vastly superior and preferable to a carbon cap-and-trade system. Here's why:

- Carbon taxes lend predictability to energy prices, whereas cap-and-trade systems
 exacerbate the price volatility that historically has discouraged investments in carbonreducing energy efficiency and carbon-replacing renewable energy.
- Carbon taxes are transparent and understandable, making them more likely to elicit
 public support than an opaque and difficult to understand cap-and-trade system. The coauthor of the U.S. Senate cap-and-trade bill, Sen. John Kerry (now Secretary of State)
 told a reporter in 2009, "I don't know what 'cap and trade' means. I don't think the
 average American does."
- Carbon taxes can be implemented more quickly than complex permit-based cap-and-trade systems.
- Carbon taxes aren't easily manipulable by special interests, whereas the complexity of cap-and-trade leaves it rife for exploitation by the financial industry.

- Carbon tax revenues can be more or less guaranteed and integrated into state or federal fiscal policy, owing to their predictability, whereas the price-volatility of cap-and-trade precludes its being counted on as a revenue source.
- Carbon taxes are replicable across borders, since the price "metric" embodied in a carbon tax is far more universal than the quantity-reduction metric underlying cap-and-trade.
- Perversely, cap-and-trade <u>discourages voluntary/individual carbon reductions</u>, since those
 cause a lowering of prices of emission permits which undercuts low-carbon investments;
 carbon taxes are free of this unintended negative consequence.

Politically, cap-and-trade has functioned as a "safe harbor" for politicians who grasp the need for pricing carbon emissions but cling to the need to "hide the price" to appease interest groups and/or ordinary citizens. But the point of carbon emissions pricing is to raise the price of emitting carbon. Better to make the price explicit, via a tax, and protect households by making the tax revenue-neutral.

CARBON TAX IN CANADA

On December 9, 2016, Prime Minister Justin Trudeau made the most important energy and environmental policy announcement in half a century with his pledge to create a national clean fuel standard "based on life cycle analysis".

The flow-through carbon tax we propose is consistent with the federal government's **National Clean Fuel Standard** policies [21]:

OBJECTIVE

The overall objective of a clean fuel standard would be to achieve annual reductions of 30 megatonnes (Mt) of GHG emissions by 2030. This reduction will provide a significant contribution towards achieving Canada's commitment of 30 percent emissions reduction below 2005 levels, by 2030. This reduction is like removing over 7-million vehicles from the roads for a year.

BROAD COVERAGE

A clean fuel standard will encourage the use of cleaner fuels in many sectors of the economy, including the fuels we use in transportation, in our homes and buildings, and the fuels that power our industries. It would address a broad suite of fuels, which could include liquid fuels (e.g., gasoline, diesel, and heavy-fuel oil), gaseous fuels (e.g., natural gas and propane), and solid fuels (e.g., petroleum coke).

PERFORMANCE-BASED APPROACH

The clean fuel standard would set requirements to reduce the lifecycle carbon intensities of fuels supplied in a given year, based on lifecycle analysis. By contrast to renewable-fuel mandates, this approach would not prescribe the particular low-carbon fuel or technology that must be used; instead, it would focus on emissions reduction. The clean fuel standard would result in decreased emissions while minimizing compliance costs.

This approach would foster the deployment of a broad range of lower-carbon fuels and alternative technologies such as electricity, biogas, hydrogen, and renewable fuels.

MEASURING CARBON INTENSITY BY LIFECYCLE ANALYSIS

Carbon-intensity standards will be set to reduce the lifecycle carbon intensities of all fuels supplied, based on lifecycle analysis. The standards could be set at the facility level, at a sector-wide average, or set on some other basis. Carbon intensity is the measure of how much carbon is emitted into the atmosphere relative to the amount of energy in the fuel consumed.

FLEXIBILITY

The standard will be designed to provide maximum flexibility to fuel suppliers, and it may include provisions to take into account regional differences, similar to those that currently exist under the Renewable Fuels Regulations.

CARBON PRICING AND CLEAN FUEL STANDARD

The clean fuel standard would complement carbon pricing by ensuring consumers have access to a suite of lower-carbon fuels. In doing so, it will also drive down emissions and

contribute to a clean-growth economy. By specifically addressing the carbon footprint of fuels, the clean fuel standard will drive innovation and create jobs and opportunities across a number of sectors, including transportation, agriculture, and clean-technology sectors.

The National Clean Fuel Standard policy mandates a broader consideration than simply transportation fuels because it covers ALL fuel uses, which represents about 80% of Canadian energy use rather than just transportation fuels as Ontario's review does. Indeed, Ontario is mandated by the Federal policy to expand their policy review to cover all fuels and to submit them and a reply to the proposed new Federal policy.

Although Ontario's Renewable Fuel Standard is applicable to transportation fuels, the Standard's mandate should be expanded to include 100% of energy production and utilization, in order not only to be compliant with the National Clean Fuel Standard policy, but to exceed it by including the 20% of energy that is electricity that the Federal government left out.

THE GREEN SHIFT - STEPHAN DION

At the federal level, a carbon tax was proposed by Stephan Dion with his Green Shift program [22]. Dion proposed a carbon tax based on the greenhouse gas emissions. It would start out at \$10/tonne, rising to \$40/tonne within 4 years and be revenue neutral through of tax credits offsetting the revenue. While not specific about the implementation, it appeared that Dion intended to replace the federal excise tax on fuel and replace it with carbon tax applied at the wholesale level. With gasoline being taxed at the equivalent of \$42/tonne of carbon dioxide, this carbon tax would not have resulted in any price increases to gasoline.

BRITISH COLUMBIA

Similar to Dion's Green Shift plan, British Columbia introduced a \$10/tonne carbon tax on Carbon Dioxide Equivalent (CO2e) emissions [23], which would rise to \$30/tonne by 2012.

BC's carbon tax was revenue neutral by reducing corporate and income taxes subject to annual legislative planning.

ALBERTA

Alberta enacted the *Specified Gas Emitters Regulation* in 2007 [24]. This carbon tax requires a \$15/tonne contribution be made to the "Climate Change and Emissions Management Fund" (CCEMF) by large carbon emitter companies (ie, emitting more than 100,000 tonnes/year of greenhouse gas) to either reduce their CO2 emissions per barrel by 12 percent, or buy an offset in Alberta to apply against their total emissions. In 2016, the contribution was raised to \$40/tonne for those large emitters. The intent of tax was to get oil companies and coal-fired generators to reduce their greenhouse gas emissions.

Starting 2017, Alberta will have a similar carbon tax scheme as British Columbia, which will be applied to the entire economy. All businesses and residents will pay a carbon tax based upon the carbon dioxide equivalent emissions, including the burning of wood and biofuels. The initial tax will be \$20 per tonne, increasing to \$30 per tonne in 2018, and increase thereafter by the rate of inflation plus 2%.

QUEBEC

Quebec enacted Canada's first carbon tax in October of 2007. It introduced a \$0.008/litre tax on diesel fuel, which the equivalent of \$3.00/tonne equivalent CO2e emissions [25]. This is not a revenue-neutral program because the revenue collected is used to fund energy-efficiency programs including public transit. Quebec's program also does not appear to be transparent because the carbon tax is not broken out as a component of Quebec's fuel taxes.

CARBON TAX IN THE USA

There is no carbon tax under consideration at the federal level in the USA. Several states have considered implementing carbon tax programs but no proposals have yet been passed into law. In addition, some municipal level jurisdictions implemented carbon taxes:

- Boulder, Colorado, November 2006
- Bay Area Air Quality Management District, which covers nine counties in the San Francisco Bay Area, May 2008
- Montgomery County, Maryland, May 2010

CARBON TAX IN OTHER COUNTRIES

The following countries have the some sort of carbon tax in place [24]. All appear to base their carbon pricing solely on carbon dioxide emissions.

- Costa Rica
- Denmark
- Finland
- France
- Germany
- Netherlands
- Norway
- Republic of Ireland
- Sweden
- Switzerland
- United Kingdom

Other Considerations

ENERGY STORAGE THROUGH VEHICLE BATTERIES

For Ontario to maximize the consumption of the renewable energy generated in the province and minimize the amount of surplus power exported to neighbouring jurisdictions, Ontario must have the ability to store renewable energy. One such means of storing electricity is through battery technology and every electric vehicle has a battery.

Most vehicles are used for commuting and are therefore parked most of the time either at work or at home. If sufficiently large enough, Ontario's entire fleet of electric vehicles could therefore be used to help balance Ontario's electricity supply when they are plugged-in so that every vehicle owner would be an IESO market participant. This could be accomplished with smart chargers that charge vehicles when there is a surplus and draw power when there is a demand. The vehicle's charge controller would ensure that the vehicle battery would not be drawn down below the necessary charge level for the homeward commute. The spread between the market price to charge the battery at low demand and the price to return electricity at high demand could help with the public's acceptance of electric vehicles.

Besides vehicle owners, IESO should allow each home or business owner to become an IESO market participant. Already, there is home battery technology available to store electricity for the home owner. In addition, other technology is being developed to store electricity (such as solid state ammonia synthesis and ammonia fuel cells) as home appliances, which would help to transform Ontario's grid from central generation to distributed generation. This would facilitate vehicle-to-grid (V2G) technology as well.

EMISSION COMPLIANCE

Greenhouse gas emissions are not limited to combustion. Many fuels themselves are greenhouse gases and fugitive emissions prior to combustion can also have significant adverse effects on global warming. It is crucial that the Ministry of Environment conduct regular inspections of fuel systems to ensure that system operators do not allow any leakage. There must be vigilance about fugitive emissions from the wellhead to the tailpipe.

Methane, for example, is emitted during the production and transport of coal, natural gas, and oil. Methane emissions also result from livestock and other agricultural practices and by the decay of organic waste in municipal solid waste landfills. According to the US EPA, 11% of greenhouse gas emissions in 2014 were from methane compared with 81% for carbon dioxide [26]. Since methane has a global warming potential of 28-36, the relatively smaller methane emission fraction had 3.8 to 4.9 times worse effect on global warming than carbon dioxide.

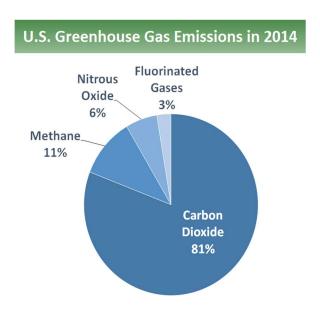


Figure 24 – U.S. Greenhouse Gas Emissions [26]

Natural gas-fueled vehicles have methane emissions both before and after combustion. In the case of LNG vehicles, as the cryogenic fuel warms, some of the methane leaks into the atmosphere as the safety relief valves blow off excess pressure. Both compression and spark ignition engines will have "methane slip" through the exhaust from incomplete natural gas combustion. Both of these sources of methane emissions must be mitigated and these vehicles should be regularly tested for emission compliance.



Figure 25 – LNG Truck Venting Excess Methane Pressure [27]

DEVELOPMENTS IN NH3 FUEL TECHNOLOGY

There has been much research that shows that ammonia can be used to fuel the most polluting forms of transport including marine, truck, and aircraft at less than half the cost and emissions of conventional fuels, even when the ammonia is made from natural gas or coal using appropriate technologies.

There have been some huge developments regarding green ammonia production, conversion of carbon in hydrocarbons into urea and use of NH3 as a fuel, for energy storage, especially in the past few months, including Mr. Gates Breakthrough Energy moving to develop technologies to produce "Zero emission ammonia", many of which actually exist commercially today.

Commercially available technologies have proven that energy and heating and cooling costs can be reduced by more than 50% when ammonia is used in the mix, and there are opportunities to use a combination of fossil fuels and electricity in much more profitable ways. There are patented engines that provide CHPC (combined heat, power and cooling) at over 60%, at a fraction of the cost of any other technologies.

Existing natural gas infrastructure and power plants can use ammonia and where NH3 plants are built beside gas plants that burn NH3 with the vast amounts of pure oxygen available results in increases in efficiency as much as 50% with similar reductions in costs and virtually zero carbon, sulfur or nitrogen oxide emitted, and nothing but nitrogen (N2) and water (H2O) created as a result.

A joint submission between *Hydrofuel Inc.* and the *University of Ontario Institute of Technology's (UOIT)* submission to the *Canadian Environmental Assessment Agency (CEAA)* for the Pacific NorthWest LNG Project concluded [28]:

"For the reasons stated in the above document and supported by Dr Ibrahim Dincer's group at UOIT (comments and MITACS report attached), Hydrofuel Inc. believes that the Pacific NorthWest LNG Project should not proceed and that the alternative of transporting energy in the form of anhydrous ammonia should be considered instead."

This submission proposes we use ammonia instead, an idea that has gained worldwide academic and public and private sector support as you will see from the details in the 13 points below and the attachments.

Hydrofuel Inc. has completed several projects with UOIT including one at the end of March 2017, which provides an analysis of the economics of converting gas into NH3 and urea for transport to worldwide markets instead of using LNG. We already know that NH3 can be burned and utilized at much higher efficiencies in engines and turbines with lower emissions, adding a price for carbon changes the numbers and adding the life cycle costs will make the economics of do so even more compelling.

In late September, 2016, Hydrofuel Inc. released the *UOIT Green NH3 and Key Life* cycle energy research [29] at the 13th NH3 Fuel Association conference at UCLA [30], where SIEMENS (supported by Innovate UK. Collaborators include the University of Oxford, Cardiff University and the Science & Technology Facilities Council) announced their "Green" Ammonia technology, pilot plant project, and strategy that "Green" ammonia is the key to meeting the twin challenges of the 21st century. [31]

There is worldwide development of dozens of NH3 production and utilization technologies being reported now, and in addition to the *NH3 Fuel Association* website [32] a new *Ammonia Energy* Global Information Portal website [33]

UOIT RESEARCH

- Key Life-Cycle Numbers for NH3, Fossil Fuel and Green Energy production and utilization in agriculture, energy and utilities, and transportation systems [34]
- Comparative Life Cycle Assessment of Various Ammonia Production Methods [35]
- The most recent UOIT/Hydrofuel Inc. peer-reviewed paper published free to read by Springer as part of their Nature SharedIt and formally in *Environmental Management* on "Impact Assessment and Environmental Evaluation of Various Ammonia Production Processes" [94]. The previous paper a "Comparative life cycle assessment of various ammonia production methods" was published in the Journal of Cleaner Power on November 1, 2016 [35].

• Other research Hydrofuel has done with UOIT may be found in the Latest News section of the Hydrofuel web site. [37]

EUROPEAN POWER TO AMMONIA CONFERENCE

1st European Power to Ammonia® Conference announced for May 18-19, 2017 - 1st European Conference on Sustainable Ammonia Solutions aims to gather thought societies, industries and academics, including well-known experts, developers and scientists to present the latest research results, present achievements, application fields and business prospects in energy solutions. This conference welcomes all researchers, industrialists, scientists as well as student and corporate delegates to participate and to have a great experience. [38]

JAPANESE AMMONIA USAGE

Japan has been using ammonia to fuel coal power plants reducing emissions. (Using similar technologies to this used in China for forty years.) Japanese utilities team on CO2-reducing tech for coal plants - Coal-ammonia fuel mix seen lowering emissions by at least 20%. [39]

DR. STEVE WITTRIG, US CLEAN AIR TASK FORCE

The development of the ammonia economy will depend on successful deployment of a range of new technologies in the context of the existing ammonia industry and the world's many existing energy markets. This report *Ammonia Fuel: Opportunities, Markets, Issues* [40] provides a framework for:

- Supply/demand and status of the global ammonia industry.
- Plausible response of growth and prices in response to a giant new market in fuels.
- New incremental applications which may lead to 'disruptive innovation' and rapid takeoff for ammonia fuels.
- Barriers to takeoff.

HYPER HUB PATENT

An improved system of hardware and controls, known as a Hyper Hub, that absorbs electric power from any source, including hydropower, wind, solar, and other renewable energy resources, chemically stores the power in hydrogen-dense anhydrous ammonia, then reshapes the stored energy to the power grid with zero emissions by using anhydrous ammonia to fuel dieseltype, spark-ignited internal combustion, combustion turbine, fuel cell or other electric power generators, and for other purposes. [41]

OTHER DEVELOPMENTS

- 1. On November 10, 2016, *Ammonia Energy* announced a 2018 pilot plant in Japan will demonstrate a new way to produce ammonia at industrial-scale, with a low carbon footprint. [42]
- 2. On November 28, 2016, the Trudeau Government announced its new life cycle Clean Fuels Standards policy being formulated for February 2017. [43]
- 3. On December 2, 2016, *Ammonia Energy* announced *Nuon*'s Power-to-Ammonia update, and the first European ammonia fuel conference in 2017. [44]
- 4. On December 6, 2016, the Massachusetts Institute of Technology (MIT) released their *PLAN B FOSSIL FUELS without CO2 emissions*. Using clean-tech and ammonia makes fossil fuels clean and much more profitable. The stored chemical potential in fossil fuels from thermonuclear derived solar energy may be utilized by less conventional means without producing carbon dioxide. Chemical pathways to produce hydrogen or ammonia from hydrocarbons without co-production of carbon dioxide are possible in new process configurations. Such processes may be more cost effective than other options and more readily implemented. [45]
- 5. Phoenix Energy, a developer of biomass gasification plants based in San Francisco, shared some interesting data today during a Webinar on "The Future of Renewable Methane in Today's Regulatory & Policy Environment." The company operates two plants in California

that together produce 5.5 tonnes/day of byproduct carbon in the form of biochar. They are currently selling all of the biochar they can produce into agricultural markets at a price of \$1,700 per tonne. [46] Unlike what many uninformed politicians, environmentalists and special interests believe or are lead to, Pheonix Energy said that their data proved:

- Ammonia as a fuel is a real interesting target, where it would allow you to make the transfer of that chemical potential to something that fits in well to our infrastructure ... It has the same heat of combustion as methanol, so it has plenty of power per weight. When you burn it though, you only make nitrogen and water. There's ways to get around the issues with NOx and stuff, so this is really a nice fuel. And the wonderful thing about ammonia, if you think about it, if you made it at a price for fuel, it would do something else that's really important for the poor, which is to lower the price of food ...
- All of the automobile companies have looked at ammonia as a possible fuel, and it's one that is certainly worthy of consideration if one wants to think about what we can do about a liquid fuel that fits into our own infrastructure that can be used more or less in our world as we know it.
- 6. On December 14, 2106, *Ammonia Energy* announced the US DOE ARPA-E's REFUEL program awarded \$35 Million (US) in funding to a total of 16 projects for carbon-neutral liquid fuels (of these 16 projects, 13 were focused on carbon free ammonia (none for hydrogen not associated with using ammonia), of the three others that went to carbon based energy, two were for Dimethyl ether, one looked at ethanol). [47]
- 7. On December 14, 2106, *Ammonia Energy* announced the Bill Gates' *Breakthrough Energy Coalition* [48] is starting work on viable grid scale and world class Green NH3 technologies after concluding that one of its initial Technical Quests is to make "Zero-GHG Ammonia Production" a reality. [49]
- 8. On December 16, 2016, the Trudeau government announced it has delayed the notification date of NRCan's decision on clean energy innovation funding proposals [50]. The date has been changed from late fall 2016 to winter (mid to late February) 2017. As Greg Vezina points out in the February 2017 Ottawa Life Magazine article, it appears the recent

developments in support of ammonia and moving away from bio-fuels and other carbon based technologies combined with the government's new Life Cycle Clean Fuels Policy has caused a major rethink of how they will spend the \$billions in proposed funding. All of this is very good news for ammonia energy proponents. However the government awarded NRCan's clean energy innovation funding grants even before they release the details of the Life Cycle Clean Fuels Policy discussion paper in late February, 2017, two months before the April 25, 2017, deadline for submissions on it, which is in fact long before they will actually know which technologies should qualify for funding. According to two of Ontario's Auditors' General the Ontario Liberal government wasted \$35 billion to date with the potential for it to rise to \$150 billion, because they did not do any analysis on their green energy policies and the awarding of subsidies. Lobbyists and political insiders were the biggest beneficiaries from these policies as 99.4 percent of the companies that got 20 year green energy subsidies contributed to the Ontario Liberal Party.

- 9. On January 27, 2017, *Ammonia Energy* announced more US DOE funding research into sustainable ammonia synthesis. These *Sustainable Ammonia Synthesis projects*, announced in August 2016 and administered by the Office of Basic Energy Sciences, aim "to investigate some of the outstanding scientific questions in the synthesis of ammonia (NH₃) from nitrogen (N₂) using processes that do not generate greenhouse gases." There are two substantial differences between these awards and the other recent DOE funding for sustainable ammonia technologies, announced in December 2016 and administered through ARPA-E's "REFUEL" program. [51]
- 10. On February 2, 2017, Ammonia Energy announced studies in both Canada and Italy including our work and a "Comparative studies of ammonia production, combining renewable hydrogen with Haber-Bosch." [52] The Japan Science and Technology Agency (JST) announced [53] that another consortium this one led by Hiroshima University and including Showa Denko, Taiyo Nichi Company, Toyota Industries and the National Institute of Advanced Industrial Science and Technology had succeeded in developing "viable technology to produce high-purity hydrogen [from an] ammonia hydrogen station."

- 11. On February 3, 2017, *Ammonia Energy* announced seven International R&D projects [54] on sustainable ammonia synthesis technologies, representing research coming out of eight countries spread across four continents. These projects span the breadth of next-generation ammonia synthesis research, from nanotechnology and electrocatalysis to plasmas and ionic liquids.
- 12. On February 9, 2017, *Ammonia Energy* announced the German Aerospace Center (DLR) at the Institute of Thermodynamic Engineering is developing a method for electrochemical ammonia synthesis [55] at ambient conditions.
- 13. On February 16, 2017, *Ammonia Energy* announced the Institute of Fluid Science at Tohoku University in Sendai, Japan, has developed the world's first technology for direct combustion of ammonia in a gas turbine [56], and a start-up company in Denmark is commercializing technology to generate low-cost, high-purity hydrogen from ammonia, for use in fuel cells [57].
- 14. On March 3, 2017, *Ammonia Energy* announced this new study [58], published in the January 2017 AIChE Journal by researchers from RWTH Aachen University and JARA-ENERGY, which examines ammonia energy storage "for integrating intermittent renewables on the utility scale," and in Israel, researchers recently published a complementary Power-to-fuel-to-power [59] study that compared carbon-based and nitrogen-based synthetic fuels.
- 15. On March 6, 2017, *Ammonia Energy* announced Nikkei Asian Review has published two articles outlining plans by major engineering and power firms to build utility-scale demonstrations using ammonia as a fuel for electricity generation [60] in ammonia-coal and ammonia-natural gas applications to reduce costs and emissions.
- 16. On March 16, 2017, *Ammonia Energy* announced a recent The Japan Times opinion piece that predicts a "revolutionary disruption coming to the energy sector," and suggests that using ammonia for energy storage will prove to be "a game-changer [61] at least on the scale of the shale oil and gas revolution."

- 17. On March 23, 2017, *Ammonia Energy* announced the shipping industry is beginning to evaluate ammonia as a potential "bunker fuel," [62] a carbon-free alternative to the heavy fuel oil (HFO) used in maritime transport.
- 18. The market for ammonia/diesel dual fuel engines in the marine industry is now quantified for 2016-2020. [63]

PART 10: Market trends

- o Ammonia gas as substitute for natural gas in dual fuel engines
- Increased OEM collaboration to study dual fuel engine technology for automobiles
- LNG: Maritime green fuel of the future
- 19. On March 30, 2017, *Ammonia Energy* announced Gifu University in Japan's breakthrough in technology for generating hydrogen from ammonia [64], and the Institute for Sustainable Process Technology's Power to Ammonia feasibility study [65] that represents a major step toward commercializing renewable ammonia and which examines the "value chains and business cases to produce CO2-free ammonia," analysing the potential for commercial deployment at three companies with existing sites in The Netherlands
- 20. On April 6, 2017, *Ammonia Energy* announced the 2017 NH3 Fuel Conference that will be staged on November 1, 2017, in conjunction with the American Institute of Chemical Engineers (AIChE) conference in Minneapolis, MN, from October 29 to November 3. The two organizations will co-sponsor a topical conference on ammonia energy called "NH3 Energy+: Enabling Optimized, Sustainable Energy and Agriculture" [66]. This year's annual AIChE meeting is expected to draw more than 8,000 attendees.
- 21. On April 20, 2017, *Ammonia Energy* announced the publication of "Auto-ignition of a carbon-free aqueous ammonia/ammonium nitrate monofuel: a thermal and barometric analysis." The authors clearly see nitrogen-based fuels as a key element of a sustainable energy future. [67]
- 22. On April 21, 2017, *Ammonia Energy* announced an important new voice joined the chorus of support for renewable ammonia and its potential use as an energy vector the International Energy Agency (IEA). Cédric Philibert, Senior Energy Analyst at the IEA,

identifies a major problem with the hydrogen economy: hydrogen is currently made from fossil fuels. But his argument for producing hydrogen from renewable energy leads almost inevitably to ammonia: "In some not-too-distant future, ammonia could be used on its own as a carbon-free fuel or as an energy carrier to store and transport energy conveniently" [68]

- 23. York University Disaster and Emergency Management degree reports (2016):
 - "Anhydrous Ammonia to Reduce Critical Infrastructure Risk" [69]
 - "Can the Smog in China be Reduced or Eliminated Using Ammonia?" [70].
- 24. Below are links to a number of articles we have written or co-written, and one that has been written about our work.
 - A. Four articles co-written for Ottawa Life Magazine.
 - i. One of Four: An Alternate View on Pipelines Transport Ammonia not Crude [71]
 - ii. Two of four: Green Ammonia [72]
 - iii. Three of four: Canada's Ammonia Energy Option [73]
 - iv. Four of Four: Canada's Life Cycle Clean Fuels Policy Game Changer [74]
 - B. Most of the information about ammonia as a renewable fuel is not covered by the mainstream media in Canada with the exception of Ottawa Life Magazine and Toronto Sun Editor Lorrie Goldstein who has asked Hydrofuel Inc. Chairman Greg Vezina to write many SUN editorial page articles on energy policy, and who has written about NH3 energy previously, most recently in:

"High Cost of Ignorance" a Toronto Sun Editorial page article on Oct. 23, 2016. [75]

25. Research on Cost comparisons of pipelines, distribution and marine transport for natural gas and ammonia. Included are <u>five</u> studies Hydrofuel Inc. have already done with UOIT on point, however, as mentioned above, MITACS Report 3 was to be released at the now

postponed US DOE ARPA-E REFUEL [58] program launch April 25-26, 2017, in Houston Texas, at which Greg Vezina was scheduled to speak about "Life cycle analysis of carbon neutral liquid fuels" on the first day and will participate in the panel discussion: *Investment in chemical energy storage/VC perspective* on the second day. [59].

- A. Ammonia as a Potential Solution for Alberta (Submitted to Alberta Premier) [76]
- B. Natural Gas to Ammonia as a Potential Solution for British Columbia (Submitted to BC Premier) [77]
- C. *MITACS Report 1* November 2015 "Comparative assessment of NH3 production and utilization in transportation systems"[78]
- D. *MITACS Report 2* June 2016 "Comparative assessment of NH3 production and utilization in agriculture, energy and utilities, and transportation systems" [79]
- E. *MITACS Report 3* March 2017 "Comprehensive Evaluation of NH3 Production and Utilization Options for Clean Energy Applications" [80]

The three MITACS Reports prove beyond a doubt the viability of the technologies UOIT and Hydrofuel Inc. have developed and demonstrated to date and the others under development.

Hydrofuel Inc. and UOIT submitted a joint report "Ammonia (NH3) as a Potential Transportation Solution for Ontario" [81] in response to the Ontario government's discussion paper "Developing a modern renewable fuel standard for gasoline in Ontario" [82] for which submissions closed on March 12, 2017.

While Ontario's review was limited to transportation fuels these comments were expanded to cover all fuel uses in accordance with the provisions outlined in the Federal government's new "Life Cycle Clean Fuels Policy" [83] review discussion paper released in late February, 2017, for which comments close April 25, 2017. The new federal standards will cover all fuels, which represents 80% of energy use, unlike Ontario's which only cover about one third of fuel use emissions, but which will none the less have to be expanded to comply with the new federal standards regardless.

Hydrofuel Inc. intends to continue its work with UOIT to complete six more projects that have many expected benefits:

- Work Package 1: to develop a system for clean ammonia production using fossil fuel, renewable and waste energy as feedstocks
- Work Package 2: to develop efficient and effective ammonia fuel cells
- Work Package 3: to improve and optimize ammonia decomposer for on-board hydrogen production and utilization in vehicles
- Work Package 4: to develop and implement the production of commercially ammonia kits for converting diesel generators to ammonia power generators
- Work Package 5: to improve and optimize ammonia heat engine for power generation and for heating and cooling applications
- Work Package 6: to study feasibility and life cycle assessments on using ammonia in various systems and applications, expanding our most recent work

These projects will help:

- to implement novel technologies for production and utilization of energy more cleanly and efficiently.
- to achieve significant emissions reductions, while maintaining Canada's economic advantage and its ability to create jobs for Canadians.
- to reduce and eliminate air pollutants from energy production and utilization.
- to contribute to the transition to a low-carbon economy and help reduce the economic, environmental, and health risks associated with air pollution and climate change.
- to significantly reduce energy use at the community level and increase its renewable content through the innovative integration of renewable energy sources for heating & cooling & electricity production and energy storage & management
- to reduce the use of diesel fuel at industrial sites through innovative alternative fuel applications
- to implement highly efficient heating and cooling technologies for micro-combined heat and power generation

For example, the retrofit technology for diesel/ammonia multifuels conversions kits for diesel engines and generators Hydrofuel Inc. have developed could be used in Ontario successfully in many applications according to an analysis in our Summary of the Proposed Project "Connection of Remote First Nation Communities in Northwest Ontario" [84]. The report proposes to use

ammonia initially shipped in and a decade later produced locally using renewable resources verses plans for a \$1.5 billion 1,500 km hydro transmission line based on a forty year business case to provide interruptible power backed up by... wait for it... diesel power.

Conclusions of UOIT's Diesel Generator Alternatives Report:

- As the global warming and energy issues become important topics to be considered for the future of remote communities, alternative solutions for power and heat generation need to be investigated.
- Currently, diesel generators running on diesel are utilized for Northwestern Ontario remote communities. This is one of the most expensive and environmentally damaging option.
- First Nations request and propose a 1,500 km power transmission line to the remote communities instead of diesel generators which is a high cost investment.
- The significance of distributed power generation is emphasized in many studies throughout the world. Hence, producing power and heat via stand-alone facilities are being encouraged by the governments and decision makers.
- As an alternative and sustainable fuel, ammonia, can be utilized in the diesel generators by
 minor modifications. Ammonia has no greenhouse gas emission during utilization in the
 diesel generators. Hence, it is the most environmentally benign fuel among other alternatives.
- Transportation and storage of ammonia is already available and well-known since it is the second largest produced chemical in the world. This implies that instead of diesel, ammonia can easily be transported and stored.
- Ammonia can be produced on-site using renewable energy resources such as wind, solar and
 hydropower which are already available in these remote communities. This brings minimum
 transportation cost. However, diesel needs to be transported for long distances.
- The operation of diesel engines with fuel ammonia has the lowest cost based on the current market prices. Considering the technology development of ammonia production, the cost of ammonia will continue to decrease which will bring additional reductions in total cost.

- Production and utilization of ammonia in diesel generators have significantly lower environmental impacts in terms of climate change and global warming.
- As a result, ammonia usage in remote communities for power and heat production will bring significant cost and environmental benefits together with public satisfaction.

There are details about and hyperlinks to much of the Hydrofuel Inc. - UOIT and other available research on the Hydrofuel Inc. website [85]. There is very solid scientific and economic research and numerous practical applications that support using ammonia to green hydrocarbons and store renewable energy already commercially available and many more emerging almost daily.

NH3 is the Obvious Carbon-Free Fuel and Energy Currency

If these technologies and opportunities are so good, why doesn't anyone know about them and why aren't the major companies in these sectors developing these technologies?

The following three examples directly on point explain the answer as succinctly as is possible under these circumstances, where there actually is a mountain of research and a century and a half of history involved to digest to quickly come up to speed on this subject matter, an experience which might otherwise be compared to drinking out of a fire hose.

EXAMPLE 1

In 1984, the National Research Council of Canada ("NRC") commissioned a report by International consulting firm Chemetics International Co. (Toronto), a division of C-I-L Inc. ("Chemetics") called, "Ammonia Vehicle Fuel - Assessment of the technical, economic and safety problems associated with ammonia use as a ground transportation fuel." [86] Among the conclusions and recommendations were:

6.0 CONCLUSIONS AND RECOMMENDATIONS

This report development started from a rather negative preconceived feeling as to the potential of ammonia as a vehicle fuel. The study has indicated a very real potential from a purely technical point of view and some positive features for safety and environmental concerns. If the economics of capital costs continue to improve and the high costs and scarcity of hydrocarbons become more acute, ammonia will become a viable option either as a carrier for hydrogen fuel or possibly as a direct combustion fuel. It is conceivable that the conditions in remote areas or in developing countries where there could be available, renewable resources to generate electricity and hence, ammonia.

6.1 CONCLUSIONS

The following listing is a generalized summarization of the conclusion that has been formed:

- 1. There are no technical or practical engineering problems that are not very logically solvable for the general usage of ammonia as an ICE fuel.
- 2. The general safety and public acceptance of ammonia as an ICE fuel should not be difficult to achieve.
- 3. There are no known ecological or environmental problems with ammonia as an ICE fuel and in fact indications are that significant improvements could be achieved over hydrocarbon fuels.
- 4. The replacement of hydrocarbon by electricity as the energy source and raw material for ammonia-fuelled ICE is technically feasible.

6.2 RECOMMENDATIONS

As the concerns for hydrocarbon conservation and environmental damage caused are on the increase, all viable options must be looked at in depth. The alternatives are many and varied and the ramifications are astronomical. It follows then that a knowledgeable planned evolution to new and probably numerous options must be set up.

No funding was provided for any ammonia fuel and energy related research in Canada for decades that followed until the Harper government funded limited research into NH3 fuel cells, renewable energy storage in ammonia and clean diesel locomotive conversion to ammonia in the GTA in the latter part of its mandate.

Example 2

We should have moved in this direction way back in 1967 when Dr. Leon Green first proposed this idea in "Energy needs versus environmental pollution: A reconciliation?" [87].

His proposal was supported by several subsequent studies [88] including in the 2012 extensive

IEEE Report by Cal Poly Tech professor William Ahlgren "The Dual-Fuel Strategy: An Energy Transition Plan," [89] the subsequent report "Planning for a hundred fold increase in ammonia production," [90] and the recently added "Prologue to the Dual Fuel Plan," by Jim Esch, IEEE Contributing Editor. [91]

In the Introduction to the *Dual Fuel Strategy* Dr. Ahlgren wrote that many of the ideas presented have a long history.

"As early as 1967 Leon Green, Jr., writing in Science magazine [1], formulated a concept for the large-scale use of ammonia as fuel. He observed: The long-term consequences of the greenhouse effect due to CO2 buildup in the atmosphere are of serious concern. . . . To remove the offending elements (carbon and sulfur) from the fuel prior to combustion is a much more efficient and less expensive procedure than trying to clean up the combustion products. ... Outlined below is a concept for energy generation in which the fossil fuels are not burned directly but serve as raw materials for the synthesis of a clean fuel. ... This clean fuel is ammonia. ... In commercial high-tonnage production of ammonia, natural gas is used as raw material for steam reforming to generate hydrogen for the synthesis reaction. In the course of this process sulfur is removed and recovered in elemental form, and CO2 is scrubbed from the stream and may be recovered for sale or use. Although current practice is to discharge this CO2 to the atmosphere, the point is that the CO2 is under control and can be condensed or caused to react so that the carbon is tied up in some useful form. ... large amounts of CO2 are recovered per unit of ammonia produced, and the commercial value of this CO2 will have a major bearing on the economic attractiveness of the concept.

Green's concept, published nearly half a century ago, is stunning in its prescience. Why has it received so little attention? Over three decades ago the U.S. Department of Energy conducted a comprehensive study of liquefied gaseous fuels, including LNG, LPG, hydrogen, and ammonia. The latter alternatives were studied because of the extreme explosion hazard associated with LNG and LPG. Bomelberg and McNaughton coauthored the report on ammonia [2], published in 1980. After a careful comparison of hydrogen and ammonia, they wrote:

It is not understandable why hydrogen as a future fuel is widely promoted, whereas ammonia is presently not considered at all. The most likely explanation appears to be that the potential use of ammonia as a substitute fuel is just too unknown, even within the technical community.

In 2012, the potential use of ammonia as fuel remains just too unknown. Why is this? Perhaps it is because the well-known hazards of ammonia cause its use as fuel to be dismissed out-of-hand. There is also a widely held (but erroneous) notion that nitrogenbased fuels must necessarily produce excessive NOx in their exhaust. These misconceptions must be dispelled; one objective of the present essay is to do so. At the same time, the validity of these concerns must be acknowledged. Ammonia is a hazardous substance, and NOx is found in the exhaust of ammonia combustion processes. We argue that ammonia can nevertheless be safely used as fuel if it is not required to serve all purposes. We outline a plan, the dual-fuel strategy that supplements ammonia with a complementary substance, methanol, well known as an alternative fuel. Ammonia and methanol each has strength that compensates the other's weakness: ammonia is carbon-free, but has high relative toxicity; 2 methanol has low relative toxicity, but contains carbon. Their use together yields a good (not perfect) solution for liquid renewable fuel, which we assert to be sine qua non for a post-petroleum global energy system. Further, we note that ammonia and methanol are but one example of such a dualfuel pair. Other, better pairs might be found, and must be sought. What we reject is hydrogen. Hydrogen is disqualified because it is a gas. In our view, the use of hydrogen as an energy vector has been thoroughly explored over a period of half a century, and has failed to come to fruition. It is now time to move on.

A purpose of this essay is to contribute to a discussion initiated in these Proceedings by Bossel [3] and continued by Abbott [4], [5]. Bossel points out the deficiencies of the hydrogen economy, and advocates an electron economy in which batteries play a central role and fuel is unnecessary. Abbott insists that fuel is essential and that the hydrogen economy is the only realistic alternative to business as done now. We agree with Bossel that hydrogen is deficient, and with Abbott that renewable fuel is indispensible. We suggest that liquid fuels, ammonia and methanol, answer many of the objections raised

by Bossel, while acknowledging Abbott's observation that renewable fuel is an essential component of any future energy strategy.

This essay is an unapologetically rhetorical work. Our purpose is to convince as many readers as possible to adopt the dual-fuel strategy and to adjust their research priorities accordingly. What the reader will find is a mixture of scientific fact, reports on demonstrated technology, speculation about what might be possible, and future-scenario narratives. This mixture is appropriate to the grand scope of our topic. The astute reader will have no difficulty identifying the character of any given statement or passage, and treating it accordingly. Further, we approach the problem at hand how best to put the global energy system on a sustainable course as one that cannot be compartmentalized. Issues of energy, environment, and economics are inextricably intertwined yet are traditionally classified in distinct intellectual categories. These three categories cannot be separated; all must be addressed simultaneously. Commercial and financial issues in particular are touched on in this essay. It is essential to do so."

Example 3

There have been a huge number of developments in technologies and applications for the production and utilization of ammonia as a fuel in the last few years and an industrial and academic stampede to demonstrate pilot projects and commercial applications.

While the history of using ammonia [92] to power engines and in solar applications dates back to the mid-1860s, long before the invention of the internal combustion engine, the reality is it has been used to power trolleys and locomotives, cars, trucks, buses, engines, generators, and even jet aircraft. Most recently it has powered hybrid electric buses in China, fuel cells in remote cellular communication towers in Africa, farm tractors and irrigation pumps in the USA, and our retrofit conversion kits to run diesel engines and generators are tested and ready for commercial demonstration.

CONCLUDING REMARKS

Ammonia is a carbon-free fuel suitable for use in the transportation sector. It has a wellestablished production and distribution infrastructure, and has zero global warming potential during operation. In addition to its attractive qualities as a fuel, ammonia is widely used as a NOx reducing agent for combustion exhaust gases using selective catalytic reduction (SCR), and its capacity as a refrigerant can be applied to recover and further utilize engine heat that would otherwise be lost. In terms of environmental sustainability, ammonia can be produced using either fossil fuels, or any renewable energy source, using heat and/or electricity, which allows for evolution of ammonia production methods and technologies in parallel with sustainable development. Ammonia as a sustainable fuel can be used in all types of combustion engines, gas turbines, burners with only small modifications and directly in fuel cells which is a very significant advantage compared to another type of fuels. Reducing the total greenhouse gas emissions from marine transportation is possible by using ammonia which is carbon-free fuel. They can be utilized for maritime ship engines directly as supplementary fuels or individual fuels. Ammonia fueled ships yield considerably lower global warming impact during operation. Ammonia as a sustainable and clean fuel in road vehicles yield also the lowest global warming potential after electric and hydrogen vehicles. As a result, ammonia usage in the communities for transportation sector will bring significant cost and environmental benefits together with public satisfaction.

You can store twice as much energy in ammonia as a liquid refrigerated to -32°F or under pressure at 250 psi as opposed to storing or transporting natural gas at much greater pressures of 5,000 psi or much lower temperatures, especially LNG at -162°C., and that the profits from using our vast natural gas reserves to make ammonia, urea (with any income from carbon black) would be five to ten times what we can get transporting LNG half way around the world, or even across the country. In terms of energy density, liquid ammonia contains 4.32 kWh per liter, which is 70% more than liquid hydrogen (2.54 kWh per liter at cryogenic temperature) or almost three times more than compressed hydrogen (1.55 kWh per liter at 700 bar).

Although stationary applications are superior and on road use may not be the most appropriate initial use of ammonia as a fuel in comparison to other fuels in terms of driving range, a 16-gallon fuel tank of ammonia provides a driving range of 470 miles, almost twice the

range of the same volume of liquid hydrogen (259 miles) and three times the range of the same volume of compressed hydrogen (158 miles).

Some of the findings of this research are astounding and such knowledge is very valuable in corporate planning over the short to long term. As you can clearly see when today tens of thousands of scientists, engineers and experts in energy, clean tech and environmental technologies in the public and private sector and academia have quickly learned of the huge existing commercial and potential future benefits related to the production and utilization of ammonia as a fuel, energy currency, for power generation and multi-purpose uses including for refrigerated transport and combined heat, power and cooling applications in commercial or residential buildings.

Most importantly, research shows the real benefits of using *Anhydrous Ammonia to Reduce Critical Infrastructure Risk* and low cost infrastructure re-purposing applications that will delay or negate the requirements to replace existing infrastructure due to incompatibility with emerging technologies, as has been clearly demonstrated with the difficulties integrating renewable energy into the power grid.

We put together this report with all the details on the state of the art of ammonia production and utilization as a fuel, energy currency and for CO2 reduction and conversions into urea, using fossil fuel, nuclear and renewable energy and waste as the feedstocks. Hydrofuel Inc. takes great pleasure to finally get to the point where may get the fair chance to prove the viability of our technologies and business plans in an emerging global market place with a level playing field.

References

- [1] Environment and Climate Change Canada, "National Inventory Report 1990-2014: Greenhouse Gas Sources and Sinks in Canada," 2016.
- [2] E. Kroch, "Ammonia-a fuel for motorbuses," *Journal of the Institute of Petroleum,* pp. 213-223, 1945.
- [3] "ASME Turbo Expo 2015: Turbine Technical Conference and Exposition," 2015.
- [4] S. Haputhanthri, T. Maxwell, J. Fleming and C. Austin, "Ammonia and Gasoline Fuel Blends for Spark Ignited Internal Combustion Engines," *Journal of Energy Resources Technology*, vol. 137, no. 6, pp. 062201-, 2015.
- [5] H. Nozari and A. Karabeyoğlu, "Numerical study of combustion characteristics of ammonia as a renewable fuel and establishment of reduced reaction mechanisms," *Fuel*, vol. 159, pp. 223-33, 2015.
- [6] A. Miller, "Ammonia Fuel For Rail Transportation," in *2006 Annual NH3 Fuel Conference*, Denver West Marriott, Golden, CO, OCTOBER 9-10, 2006.
- [7] A. Reiter and S.-C. Kong, "Combustion and emissions characteristics of compression-ignition engine using dual ammonia-diesel fuel," *Fuel*, vol. 90, no. 1, pp. 87-97, 2011.
- [8] K. Ryu, G. Zacharakis-Jutz and S.-C. Kong, "Performance enhancement of ammonia-fueled engine by using dissociation catalyst for hydrogen generation," *International Journal of Hydrogen Energy*, vol. 39, no. 5, pp. 2390-8, 2014.
- [9] C. Zamfirescu and I. Dincer, "Ammonia as a green fuel and hydrogen source for vehicular applications," *Fuel Processing Technology*, vol. 90, no. 5, pp. 729-37, 2009.
- [10] C. Zamfirescu and I. Dincer, "Using ammonia as a sustainable fuel," *Journal of Power Sources,* vol. 185, no. 1, pp. 459-65, 2008.
- [11] B. Hinnemann and J. Nørskov, "Catalysis by Enzymes: The Biological Ammonia Synthesis," *Topics in Catalysis*, vol. 37, no. 1, pp. 55-70, 2006.
- [12] R. Lan, J. Irvine and S. Tao, "Synthesis of ammonia directly from air and water at ambient temperature and pressure," *Scientific Reports*, vol. 3, no. 1145, 2013.
- [13] F.-F. Li and S. Licht, "Advances in Understanding the Mechanism and Improved Stability of the Synthesis of Ammonia from Air and Water in Hydroxide Suspensions of Nanoscale Fe2O3," *Inorganic Chemistry*, vol. 53, no. 19, pp. 10042-4, 2014.
- [14] S. Licht, B. Cui, B. Wang, F.-F. Li, J. Lau and S. Liu, "Ammonia synthesis by N2 and steam electrolysis

- in molten hydroxide suspensions of nanoscale Fe2O3," *Science*, vol. 345, no. 6197, pp. 637-40, 2014.
- [15] T. Paschkewitz, Ammonia Production at Ambient Temperature and Pressure: An Electrochemical and Biological Approach., University of Iowa, 2012.
- [16] A. Skodra and M. Stoukides, "Electrocatalytic synthesis of ammonia from steam and nitrogen at atmospheric pressure," *Solid State Ionics*, vol. 180, no. 23-25, pp. 1332-6, 2009.
- [17] "Facts and Figures," Air Transport Action Group, May 2016. [Online]. Available: http://www.atag.org/facts-and-figures.html.
- [18] "Understanding Global Warming Potentials," 23 02 2017. [Online]. Available: https://www.epa.gov/ghgemissions/understanding-global-warming-potentials.
- [19] "Terra preta," 23 02 2017. [Online]. Available: https://en.wikipedia.org/wiki/Terra_preta.
- [20] Cap-and-Trade's Inherent Defects, "Cap-and-Trade's Inherent Defects," 24 02 2017. [Online]. Available: https://www.carbontax.org/cap-and-trade-problems/.
- [21] "Government of Canada to work with provinces, territories, and stakeholders to develop a clean fuel standard," 25 11 2016. [Online]. Available: http://news.gc.ca/web/article-en.do?nid=1160579.
- [22] S. Dion, "the Green Shift," 23 02 2017. [Online]. Available: https://www.poltext.org/sites/poltext.org/files/plateformes/ca2008lib_plt_eng._05012009_11161 7.pdf.
- [23] "British Columbia carbon tax," 23 02 2017. [Online]. Available: https://en.wikipedia.org/wiki/British_Columbia_carbon_tax.
- [24] Wikipedia, "Carbon Tax," 23 02 2017. [Online]. Available: https://en.wikipedia.org/wiki/Carbon_tax.
- [25] "Quebec's Green Future: The Lowest-Cost Route to Greenhouse Gas Reductions," 01 10 2009.
 [Online]. Available:
 https://www.cdhowe.org/sites/default/files/attachments/research_papers/mixed//backgrounder
 _118_English.pdf.
- [26] "Overview of Greenhouse Gases," 23 02 2017. [Online]. Available: https://www.epa.gov/ghgemissions/overview-greenhouse-gases.
- [27] Ultra Blaze, "Venting LNG Tanks," YouTube, 02 11 2015. [Online]. Available: https://youtu.be/vT4nG-Z3zdE.
- [28] Hydrofuel Inc. & UOIT, "Pacific NorthWest LNG Project," 16 03 2016. [Online]. Available: http://www.ceaa-acee.gc.ca/050/documents/p80032/108890E.pdf. [Accessed 08 03 2017].

- [29] Hydrofuel Inc & UOIT, "Key Life-Cycle Numbers for NH3 and other fuels," 16 09 2016. [Online]. Available: http://www.marketwired.com/press-release/new-evidence-on-green-ammonia-energy-solutions-be-presented-los-angeles-conference-2159195.htm. [Accessed 08 03 2017].
- [30] NH3 Fuel Association, "NH3 FUEL CONFERENCE 2016: SCHEDULE," [Online]. Available: https://nh3fuelassociation.org/events-conferences/conference2016/#schedule. [Accessed 08 03 2017].
- [31] Siemens PLC, ""Green" ammonia is the key to meeting the twin challenges of the 21st century.," [Online]. Available: https://www.siemens.co.uk/en/insights/potential-of-green-ammonia-asfertiliser-and-electricity-storage.htm. [Accessed 08 03 2017].
- [32] NH3 Fuel Association, "NH3 Fuel Association," [Online]. Available: http://nh3fuelassociation.org/. [Accessed 08 03 2017].
- [33] Ammonia Energy, "The global information portal for ammonia as an energy vector," [Online]. Available: http://www.ammoniaenergy.org/. [Accessed 08 03 2017].
- [34] Hydrofuel Inc. & UOIT, "Key Life-Cycle Numbers for NH3, Fossil Fuel and Green Energy production and utilization in agriculture, energy and utilities, and transportation systems," [Online]. Available: https://www.academia.edu/28681084/Key_Life-Cycle_Numbers_for_NH3_Fossil_Fuel_and_Green_Energy_production_and_utilization_in_agricult ure_energy_and_utilities_and_transportation_systems. [Accessed 08 03 2017].
- [35] Hydrofuel Inc. & UOIT, "Comparative Life Cycle Assessment of Various Ammonia Production Methods," 01 11 2016. [Online]. Available: https://www.academia.edu/19786635/Comparative_Life_Cycle_Assessment_of_Various_Ammonia_Production_Methods. [Accessed 08 03 2017].
- [36] Hydrofuel Inc., "Hydrofuel Inc. Latest News," [Online]. Available: http://nh3fuel.com/index.php?option=com_content&task=category§ionid=1&id=1&Itemid=36 . [Accessed 08 03 2017].
- [37] Proton Ventures, "NH3 Event The Biggest Ammonia Event in Europe," [Online]. Available: http://nh3event.com/. [Accessed 08 03 2017].
- [38] Nikkei Asian Review, "Japanese utilities team on CO2-reducing tech for coal plants," 02 03 2017. [Online]. Available: http://asia.nikkei.com/Business/Deals/Japanese-utilities-team-on-CO2-reducing-tech-for-coal-plants?n_cid=NARAN1507. [Accessed 08 03 2017].
- [39] D. S. Wittrig, "Clean Air Task Force Ammonia Fuel: Opportunities, Markets, Issues," [Online]. Available: https://arpa-e.energy.gov/sites/default/files/Wittrig_Ammonia_TransportationFuels_Workshop.pdf. [Accessed 08 03 2017].
- [40] J. S. Robertson, "Energy conversion system," Google Patents, 26 09 2013. [Online]. Available:

- https://www.google.com/patents/US20130252120. [Accessed 08 03 2017].
- [41] T. Brown, "Low-carbon ammonia synthesis: Japan's 'Energy Carriers'," Ammonia Industry, 10 11 2016. [Online]. Available: http://www.ammoniaenergy.org/low-carbon-ammonia-synthesis-japans-energy-carriers/. [Accessed 08 03 2017].
- [42] Canadian Manufacturing, "Feds pledge to create national clean fuel standard in latest piece of climate plan," 28 11 2016. [Online]. Available:

 http://www.canadianmanufacturing.com/environment-and-safety/feds-pledge-to-create-national-clean-fuel-standard-in-latest-piece-of-climate-plan-179601/. [Accessed 08 03 2017].
- [43] T. Brown, "Nuon's Power-to-Ammonia update, and the first European ammonia fuel conference in 2017," Ammonia Industry, 02 12 2016. [Online]. Available: http://www.ammoniaenergy.org/nuons-power-to-ammonia-update-and-the-first-european-ammonia-fuel-conference-in-2017/. [Accessed 08 03 2017].
- [44] MIT Energy Initiative, "Plan B: Fossil fuels without CO2," 06 12 2016. [Online]. Available: http://energy.mit.edu/news/plan-b-fossil-fuels-without-co2/. [Accessed 08 03 2017].
- [45] S. Crolius, "Methane to Ammonia via Pyrolysis," Ammonia Energy, 26 01 2017. [Online]. Available: http://www.ammoniaenergy.org/methane-to-ammonia-via-pyrolysis/. [Accessed 08 03 2017].
- [46] T. Brown, "ARPA-E funding for renewable ammonia synthesis technologies," Ammonia Industry, 22 12 2016. [Online]. Available: https://ammoniaindustry.com/arpa-e-funding-for-renewable-ammonia-synthesis-technologies/. [Accessed 08 03 2017].
- [47] Breakthrough Energy, "Breakthrough Energy Coalition," [Online]. Available: http://www.b-t.energy/coalition/. [Accessed 08 03 2017].
- [48] T. Brown, "Breakthrough Energy Coalition targets carbon-free ammonia," Ammonia Industry, 14 12 2016. [Online]. Available: https://ammoniaindustry.com/breakthrough-energy-coalition-targets-carbon-free-ammonia/. [Accessed 08 03 2017].
- [49] Natural Resources Canada, "The Energy Innovation Program: Frequently Asked Questions," 20 10 2016. [Online]. Available: http://www.nrcan.gc.ca/energy/science/programs-funding/18384#li-15. [Accessed 08 03 2017].
- [50] T. Brown, "US DOE funding research into sustainable ammonia synthesis," Ammonia Industry, 27 01 2017. [Online]. Available: https://ammoniaindustry.com/us-doe-funding-research-into-sustainable-ammonia-synthesis/. [Accessed 08 03 2017].
- [51] T. Brown, "Comparative studies of ammonia production, combining renewable hydrogen with Haber-Bosch," Ammonia Energy, 02 02 2017. [Online]. Available: http://www.ammoniaenergy.org/comparative-studies-of-ammonia-production-combining-renewable-hydrogen-with-haber-bosch/. [Accessed 21 04 2017].

- [52] Hiroshima University Showa Denko KK Taiyo Nichi Co., Ltd. Science and Technology Promotion Organization (JST), "Manufacture hydrogen fuel for fuel cell vehicles from ammonia," 19 07 2016. [Online]. Available: http://www.jst.go.jp/pr/announce/20160719-2/#YOUGO3. [Accessed 21 04 2017].
- [53] T. Brown, "International R&D on sustainable ammonia synthesis technologies," Ammonia Industry, 03 02 2017. [Online]. Available: https://ammoniaindustry.com/international-rd-on-sustainable-ammonia-synthesis-technologies/. [Accessed 21 04 2017].
- [54] T. Brown, "On the Ground in Germany," Ammonia Energy, 09 02 2017. [Online]. Available: http://www.ammoniaenergy.org/renewable-ammonia-in-germany/. [Accessed 21 04 2017].
- [55] T. Brown, "Ammonia-Fueled Gas Turbine Power Generation," Ammonia Energy, 16 02 2017. [Online]. Available: http://www.ammoniaenergy.org/ammonia-fueled-gas-turbine-powergeneration/. [Accessed 21 04 2017].
- [56] T. Brown, "Ammonia cracking to high-purity hydrogen, for PEM fuel cells in Denmark," Ammonia Energy, 16 02 2017. [Online]. Available: http://www.ammoniaenergy.org/ammonia-cracking-to-high-purity-hydrogen-for-pem-fuel-cells-in-denmark/. [Accessed 21 04 2017].
- [57] G. Wang, A. Mitsos and W. Marquardt, "Conceptual design of ammonia-based energy storage system: System design and time-invariant performance," *AIChE J*, vol. 63, no. 5, p. 1620–1637, 2017.
- [58] T. Brown, "Ammonia for energy storage: economic and technical analysis," Ammonia Industry, 02 03 2017. [Online]. Available: https://ammoniaindustry.com/ammonia-for-energy-storage-economic-and-technical-analysis/. [Accessed 21 04 2017].
- [59] T. Brown, "Industrial demonstrations of ammonia fuel in Japan," Ammonia Industry, 06 03 2017. [Online]. Available: https://ammoniaindustry.com/industrial-demonstrations-of-ammonia-fuel-in-japan/. [Accessed 21 04 2017].
- [60] T. Brown, "Ammonia for energy storage: a "revolutionary disruption"," Ammonia Energy, 16 03 2017. [Online]. Available: http://www.ammoniaenergy.org/ammonia-for-energy-storage-a-revolutionary-disruption/. [Accessed 21 04 2017].
- [61] T. Brown, "Bunker Ammonia: carbon-free liquid fuel for ships," Ammonia Energy, 23 03 2017. [Online]. Available: http://www.ammoniaenergy.org/bunker-ammonia-carbon-free-liquid-fuel-for-ships/. [Accessed 21 04 2017].
- [62] "Global Dual Fuel Engine Market 2016-2020," Technavio Discover Market Opportunities, 07 2016.
 [Online]. Available: https://www.technavio.com/report/global-tools-and-components-dual-fuel-engine-market. [Accessed 21 04 2017].
- [63] T. Brown, "Power to Ammonia feasibility study," Ammonia Energy, 30 03 2017. [Online]. Available: http://www.ammoniaenergy.org/power-to-ammonia-feasibility-study/. [Accessed 21 04 2017].

- [64] S. Crolius, "NH Energy+: Topical Conference at AIChE Annual Meeting," Ammonia Energy, 06 04 2017. [Online]. Available: http://www.ammoniaenergy.org/nh-energy-topical-conference-at-aiche-annual-meeting/. [Accessed 21 04 2017].
- [65] S. Crolius, "Ammonia and Other Nitrogen-Based Fuels," Ammonia Energy, 20 04 2017. [Online]. Available: http://www.ammoniaenergy.org/ammonia-and-other-nitrogen-based-fuels/. [Accessed 21 04 2017].
- [66] "REFUEL: Renewable Energy to Fuels Through Utilization of Energy-Dense Liquids," U.S. Department of Energy, 15 12 2016. [Online]. Available: https://arpa-e.energy.gov/?q=arpa-e-programs/refuel. [Accessed 21 04 2017].
- [67] "ARPA-E REFUEL Kickoff Meeting, Houston, TX 2017," US Department of Energy, 16 03 2017.
 [Online]. Available:
 https://custom.cvent.com/AA09371532024F85AB8D6B671E9321D5/files/1d76b74231dd4941832f
 d4489b3ce6fc.docx. [Accessed 21 04 2017].
- [68] Hydrofuel Inc. & UOIT, "Impact Assessment and Environmental Evaluation of Various Ammonia Production Processes," [Online]. Available: http://link.springer.com/article/10.1007%2Fs00267-017-0831-6. [Accessed 08 03 2017].
- [69] NH3 Fuel Association, "Comparisons The Logical Path Forward," [Online]. Available: https://nh3fuelassociation.org/comparisons/. [Accessed 08 03 2017].
- [70] G. Vezina and F. Raso, "An Alternate View on Pipelines Transport Ammonia not Crude," Ottawa Life Magazine, 11 07 2016. [Online]. Available: http://www.ottawalife.com/article/an-alternate-view-on-pipelines-transport-ammonia-not-crude. [Accessed 08 03 2017].
- [71] G. Vezina and I. Dincer, "Green Ammonia," Ottawa Life Magazine, 16 09 2016. [Online]. Available: http://www.ottawalife.com/article/green-ammonia. [Accessed 08 03 2017].
- [72] G. Vezina and S. Wittrig, "Canada's Ammonia Energy Option," Ottawa Life Magazine, 15 11 2016. [Online]. Available: http://www.ottawalife.com/article/canada-s-ammonia-energy-option?c=9. [Accessed 08 03 2017].
- [73] G. Vezina and F. Raso, "Canada's Life Cycle Clean Fuels Policy Game Changer," Ottawa Life Magazine, 23 01 2017. [Online]. Available: http://www.ottawalife.com/article/canada-s-life-cycle-clean-fuels-policy-game-changer. [Accessed 08 03 2017].
- [74] L. Goldstein, "High cost of ignorance," Toronto Sun, 22 10 2016. [Online]. Available: http://www.torontosun.com/2016/10/21/high-cost-of-ignorance. [Accessed 08 03 2017].
- [75] Hydrofuel Inc. & UOIT, "Ammonia as a Potential Solution for Alberta," 07 08 2016. [Online]. Available: http://nh3fuel.com/images/documents/2016-08-07%20- %20Ammonia%20Production%20for%20Alberta.pdf. [Accessed 08 03 2017].

- [76] Hydrofuel Inc. & UOIT, "Natural Gas to Ammonia as a Potential Solution for British Columbia," 09 08 2016. [Online]. Available: http://nh3fuel.com/images/documents/2016-08-09%20-%20Ammonia%20Instead%20of%20Natural%20Gas%20for%20BC.pdf. [Accessed 08 03 2017].
- [77] Hydrofuel Inc. & UOIT, "Comparative assessment of NH3 production and utilization in agriculture, energy and utilities, and transportation systems," 17 06 2016. [Online]. Available: http://nh3fuel.com/images/documents/2016-06-17%20-%20MITACS-Final%20Report-P2.pdf. [Accessed 08 03 2017].
- [78] G. Vezina, "Editorial Article Author Greg Vezina," Toronto Sun, [Online]. Available: http://www.torontosun.com/search?cx=016362519718727753442:ubvjwz0hpe0&cof=FORID:11&i e=UTF-8&q=%22Greg+Vezina%22&siteurl=www.torontosun.com/&ref=&ss=1514j2292196j2. [Accessed 08 03 2017].
- [79] I. Dincer and Y. Bicer, "Comprehensive Evaluation of NH," Hydrofuel Inc. & UOIT, 25 03 2017. [Online]. Available: http://www.nh3fuel.com/images/documents/2017-03-25%20-%20MITACS-Final-Report-P3-IT08015.pdf. [Accessed 21 04 2017].
- [80] Y. Bicer, I. Dincer, G. Vezina and F. Raso, "Impact Assessment and Environmental Evaluation of Various Ammonia Production Processes," Hydrofuel Inc. & UOIT, 14 02 2017. [Online]. Available: http://www.readcube.com/articles/10.1007/s00267-017-0831-6?author_access_token=MYnucYZr5t7vIs0I-yGtZ_e4RwIQNchNByi7wbcMAY4AiStKfv68gRrCJvRKLH6G6FVerSnoUyRDjpAoU_jbEHe-5vxAQg92U9ydQMSO45LFpCYfUdW7efZDV-G7XbrDCJmgMFYc22Ca2CJ_zSZ-8A%3D%3D. [Accessed 21 04 2017].
- [81] I. Dincer, Y. Bicer, G. Vezina and F. Raso, "Ammonia (NH3) as a Potential Transportation Solution for Ontario," Hydrofuel & UOIT, 10 03 2017. [Online]. Available: http://www.nh3fuel.com/images/documents/2017-03-10%20-%20Ontario-RFS-Discussion-Paper.pdf. [Accessed 21 04 2017].
- [82] Ontario, Environmental Registry, "Discussion Paper: Developing a Modern Renewable Fuel Standard For Gasoline in Ontario," 11 01 2017. [Online]. Available: https://www.ebr.gov.on.ca/ERS-WEB-External/displaynoticecontent.do?noticeId=MTI4OTAz&statusId=MTk0OTQ4. [Accessed 21 04 2017].
- [83] "Clean Fuel Standard: Discussion Paper," Environment and Climiate Change Canada, 06 03 2017. [Online]. Available: http://ec.gc.ca/lcpe-cepa/default.asp?lang=En&xml=D7C913BB-13D0-42AF-9BC7-FBC1580C2F4B. [Accessed 21 04 2017].
- [84] Hydrofuel Inc., "Hydrofuel Inc.," [Online]. Available: http://nh3fuel.com/. [Accessed 21 04 2017].
- [85] R. Knowlton and H. Wright, "Ammonia Vehicle Fuel Assessment of the technical, economic and safety problems associated with ammonia use as a ground transportation fuel," Chemetics International Company, 04 1984. [Online]. Available:

- http://www.nh3fuel.com/images/documents/1984-04%20-%20Chemetics%20NCR%20CAN.pdf. [Accessed 21 04 2017].
- [86] L. J. Green, "Energy Needs versus Environmental Pollution: A Reconciliation?," *American Association for the Advancement of Science*, vol. 156, no. 3781, pp. 1448-1450, 16 06 1967.
- [87] W. Ahlgren, "William Ahlgren comment to AB1257 Natural Gas Act Report," California Polytechnic State University, 04 05 2015. [Online]. Available: http://docketpublic.energy.ca.gov/PublicDocuments/15-IEPR-04/TN204490_20150504T125450_William_Ahlgren_comment_to_AB1257_Natural_Gas_Act_Report.pdf. [Accessed 21 04 2017].
- [88] W. Ahlgren, "The Dual-Fuel Strategy: An Energy Transition Plan," Proceedings of the IEEE, 11 2012. [Online]. Available: http://www.nh3fuel.com/images/documents/IEEE%20-%20Dual%20Fuel%20Strategy.pdf. [Accessed 21 04 2017].
- [89] W. Ahlgren, "Planning for Hundred-Fold Increase," California Polytechnic State University, San Luis Obispo, 2013. [Online]. Available: http://www.nh3fuel.com/images/documents/Planning%20for%20Hundred-Fold%20Increase.pdf. [Accessed 21 04 2017].
- [90] J. Esch, "Prolog to "The Dual-Fuel Strategy: An Energy Transition Plan'," IEEE, 16 10 2012. [Online]. Available: http://ieeexplore.ieee.org/document/6331607/. [Accessed 21 04 2017].
- [91] "What are some surprising details about using ammonia as a fuel?," Hydrofuel Inc., [Online]. Available: http://nh3fuel.com/index.php?option=com_content&task=view&id=37&Itemid=37. [Accessed 21 04 2017].
- [92] S. Crolius, "New Technology for Generating Hydrogen from Ammonia," Ammonia Energy, 30 03 2017. [Online]. Available: http://www.ammoniaenergy.org/new-technology-for-generating-hydrogen-from-ammonia/. [Accessed 21 04 201].
- [93] Hydrofuel Inc. & UOIT, "Summary of the Proposed Project "Connection of Remote First Nation Communities in Northwest Ontario"," [Online]. Available: http://nh3fuel.com/images/documents/Diesel%20Generator%20Alternatives.pdf. [Accessed 21 04 2017].
- [94] A. Vezina, "Anhydrous Ammonia to Reduce Critical Infrastructure Risk," York University, [Online]. Available: http://www.nh3fuel.com/images/documents/A-Vezina%20-%20Anhydrous%20Ammonia%20to%20Reduce%20Critical%20Infrastructure%20Risk.pdf. [Accessed 21 04 2017].
- [95] A. Vezina, "Can the Smog in China be Reduced or Eliminated Using Ammonia?," York University, [Online]. Available: http://www.nh3fuel.com/images/documents/A-Vezina%20-%20China%20Ammonia%20Research%20Final.pdf. [Accessed 21 04 2017].

- [96] T. Brown, "IEA calls for renewable hydrogen and carbon-free ammonia," Ammonia Energy, 21 04 2017. [Online]. Available: http://www.ammoniaenergy.org/iea-calls-for-renewable-hydrogen-and-carbon-free-ammonia/. [Accessed 21 04 2017].
- [97] Hydrofuel Inc. & UOIT, "Comparative assessment of NH3 production and utilization in transportation systems for Ontario," 23 11 2015. [Online]. Available: http://nh3fuel.com/images/documents/2015-11-23%20-%20MITACS-Final-Report-P1.pdf. [Accessed 08 03 2017].